Chapter 9 Physical Science Study Guide:

- Know all vocab (on Quizlet)
- Know/describe the different parts of an atom
- Describe the atomic model timeline
 - → Scientists, theories, discoveries, models, experiments
- Know what quarks are
- Know what happens to a neutral atom when the number of protons, neutrons, or electrons change
- · Be able to describe the three types of nuclear decay

Different Parts of an Atom:

- Electrons are very small particles inside an atom with one negative charge (-1). The closer electrons are to the nucleus, less energy.
- The nucleus is a small area in the center of the atom where most of its mass and positive charge is concentrated
- A proton is an atomic particle that has one positive charge (+1)
- A neutron is a neutral particle that exists in the nucleus of an atom
- An electron cloud is an area around an atomic nucleus where an electron is most likely to be located
- A quark is the smallest particle in an atom, and it is what protons and neutrons are made of

The Atomic Model Timeline:

DEMOCRITUS:

- He was a philosopher who challenged the popular idea of matter.
- His atomic theories were that atoms are small solid objects that cannot be divided, created, or destroyed. Also, that atoms are constantly moving in empty space, and that different types of matter are made of different types of atoms. His last theory is that the properties of the atoms determine the properties of matter.

DALTON:

- In the late 1700s, John Dalton revisited the idea of atoms.
- His atomic theories were that all matter is made of atoms that can't be divided, created, or destroyed. Also, that during a chemical reaction, atoms of one element can't be converted into atoms of another element. The third theory is that atoms of one element are identical to each other but different from atoms of another element. The last theory is that atoms combine in specific ratios.

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Kevin Wilhelm, Peter A. Soyka, Eric Olson

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Passing the State Science Proficiency Tests Craig A. Wilson, 2013-12-05 Passing the FCS physical science L2,2007 State Science Proficiency Tests presents essential content for elementary and middle school teachers who want to improve their science content background enhance their classroom instruction or pass the state science proficiency tests This book addresses different aspects of the physical life and earth sciences Each chapter was written by a science education expert and includes review questions with an accompanying answer key This book will enhance the effectiveness and competency of any pre service or in service elementary or middle school teacher Physical Science Anted, Grades 6-9 David W. Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Appenbrink, 1986 Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Chemical News and Journal of Physical Science** .1920 Chemical News and Journal of Physical Science William Crookes, James H. Gardiner, Gerald Druce, H. W. The Chemical News and Journal of Physical Science ,1920 Blood-Ryan, 1920 FCS Physical Science L3 Karen Morrison, 2008 Focus on Physical Science California Edition Michael J. Padilla, 2007 **Young Scientist Series ICSE Physics 6**, **Probability and Statistics for Physical Sciences** Brian Martin, Mark Hurwitz, 2023-09-05 Probability and Statistics for Physical Sciences Second Edition is an accessible guide to commonly used concepts and methods in statistical analysis used in the physical sciences This brief yet systematic introduction explains the origin of key techniques providing mathematical background and useful formulas The text does not assume any background in statistics and is appropriate for a wide variety of readers from first year undergraduate students to working scientists across many disciplines Provides a collection of useful formulas with mathematical background Includes worked examples throughout and end of chapter problems for practice Offers a logical progression through topics and methods in statistics and probability

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