

Ideal Gas Law Worksheet PV = nRT

Use the ideal gas law, "PerV-nRT", and the universal gas constant $R = 0.0821 \frac{L*atm}{L*mol}$ to solve the following problems:

If pressure is needed in kPa then convert by multiplying by 101.3kPa / latm to get

R =8.31 kPa*L / (K*mole)

1) If I have 4 moles of a gas at a pressure of 5.6 atm and a volume of 12 liters, what is the temperature?

2) If I have an unknown quantity of gas at a pressure of 1.2 atm, a volume of 31 liters, and a temperature of 87 °C, how many moles of gas do I have?

3) If I contain 3 moles of gas in a container with a volume of 60 liters and at a temperature of 400 K, what is the pressure inside the container?

4) If I have 7.7 moles of gas at a pressure of 0.09 atm and at a temperature of 56 °C, what is the volume of the container that the gas is in?

5) If I have 17 moles of gas at a temperature of 67 °C, and a volume of 88.89 liters, what is the pressure of the gas?

6) If I have an unknown quantity of gas at a pressure of 0.5 atm, a volume of 25 liters, and a temperature of 300 K, how many moles of gas do I have?

Pearson Prentice Hall Answer Key Ideal Gases

Holly Moore

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Elementary Principles of Chemical Processes Richard M. Felder, Ronald W. Rousseau, Lisa G. Bullard, 2020-08-11 This best selling text prepares students to formulate and solve material and energy balances in chemical process systems and lays the foundation for subsequent courses in chemical engineering The text provides a realistic informative and positive introduction to the practice of chemical engineering Applied Petroleum Reservoir Engineering Ronald E. Terry, J. Brandon Rogers, Benjamin Cole Craft, 2015 This book presents many real field examples demonstrating the use of material balance and history matching to predict reservoir performance For the first time this edition uses Microsoft Excel with VBA as its calculation tool making calculations far easier and more intuitive for today s readers Beginning with an introduction of key terms detailed coverage of the material balance approach and progressing through the principles of fluid flow water influx and advanced recovery techniques this book will be an asset to students without prior exposure to petroleum engineering with this text updated to reflect modern industrial practice

The Software Encyclopedia ,1988

CFD. James O. Wilkes, 2006 This second edition contains extensive new coverage of both microfluidics and computational fluid dynamics systematically demonstrating CFD through detailed examples using FlowLab and COMSOL Multiphysics The chapter on turbulence has been extensively revised to address more complex and realistic challenges including turbulent mixing and recirculating flows Jacket Petroleum Production Systems Michael J. Economides, 2013 Written by four leading experts this edition thoroughly introduces today s modern principles of petroleum production systems development and operation considering the combined behaviour of reservoirs surface equipment pipeline systems and storage facilities. The authors address key issues including artificial lift well diagnosis matrix stimulation hydraulic fracturing and sand control. They show how to optimise systems for diverse production schedules using queuing theory as well as linear and dynamic programming Throughout they provide both best practices and rationales fully illuminating the exploitation of unconventional oil and gas reservoirs Updates include Extensive new coverage of hydraulic fracturing including high permeability fracturing New sand and water management techniques. An all new chapter on Production Analysis New

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M. Hata,2008 The approach taken in this book is to approach vacuum systems from a pressure regime viewpoint That is after covering some basic chemistry the first pressure regime covered is the rough vacuum regime Within the study of rough vacuum systems the following topics are covered the gas load the pumping mechanism pressure measurement and vacuum system construction The discussion of rough vacuum is then followed by the study of high vacuum systems The same topics are revisited but this time from a high vacuum perspective Once both rough vacuum and high vacuum systems are covered then the topics of leak detection and residual gas analysis are introduced This approach lends itself to laboratory experimentation During the review of gas laws from chemistry there are a number of experiments and demonstrations that can be performed to reinforce basic laws and concepts Then during the study of rough vacuum systems pumpdown times can be calculated and pumpdowns performed in the laboratory Likewise during the study of high vacuum systems pumpdowns as well as other lab exercises such as outgassing and residual gas analysis can be conducted Coverage of vacuum systems from a pressure regime viewpoint rather than from a topical viewpoint Focus on the fundamental science behind vacumm system components This book is written for users of vacuum systems especially the technicians that are responsible for maintaining

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