

## Chemistry Review – Answer Key

### Chemistry Review

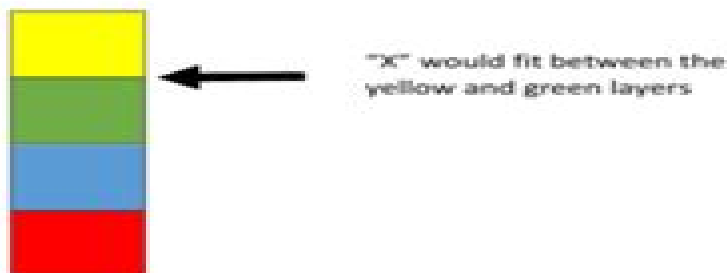
1. Identify and describe intrinsic properties of matter. Include the definition for intrinsic property and at least 2 examples.

- Intrinsic properties are either physical or chemical properties of a substance. They are properties that do not change even if the size of the substance changes.

Examples:

Physical – no new substance is formed	Chemical – new substances are formed
Density	Rusting
Magnetism	Burning
Melting/boiling temperature	Exploding
Solubility (dissolves in water)	Reacts with acid to form bubbles
Conducts (moves) electricity	

2. You have 4 liquids (Red, yellow, blue and green). Red has a density of 2.0, yellow has a density of 0.4, blue has a density of 1.0, and green has a density of 0.8. Sort the liquids according to density. A 5<sup>th</sup> liquid, "X" has a density of 0.6 than blue, but more dense than yellow. Where would liquid "X" fit?



3. You have 3 liquids: water, glycerin, and oil. Glycerin is more dense than water, and oil is less dense than water. If mixed together, in what order with these liquids be layered in a cylinder?

Oil – least dense
Water
Glycerin – most dense

4. You are given a white mystery substance that you think is either white sand or salt. What property could you use to identify your substance? Explain how you would test this.
- Solubility (dissolves in water). Salt would dissolve, sand would not.

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