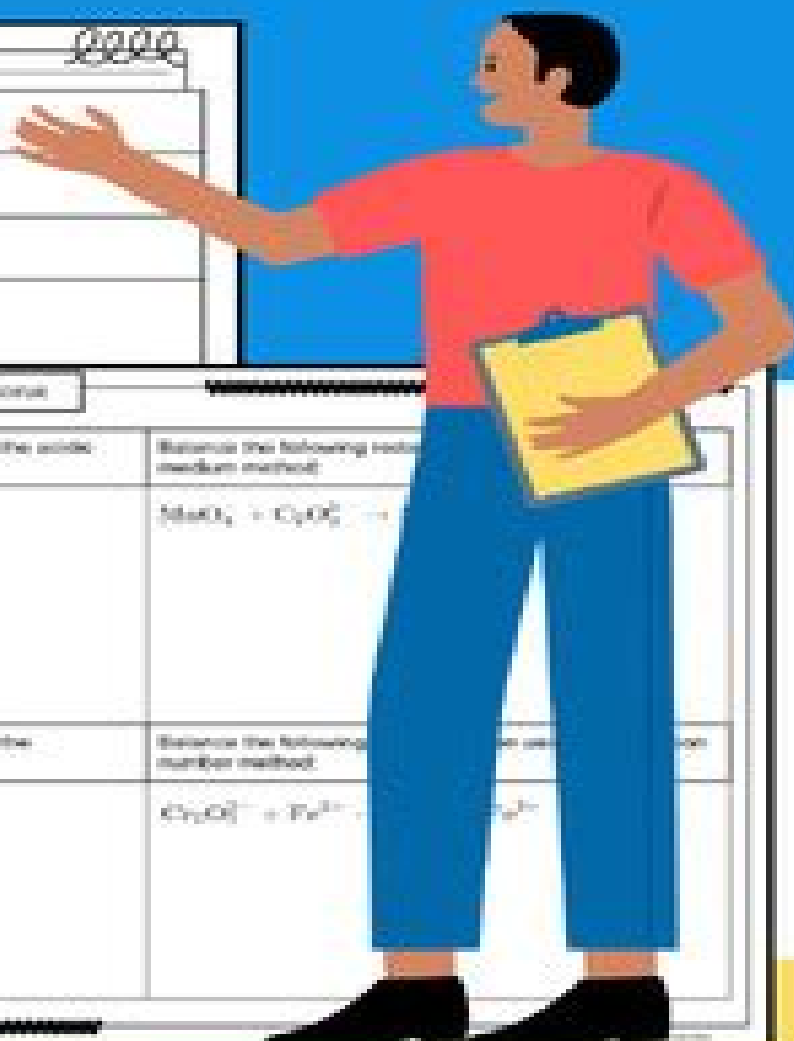


OXIDATION-REDUCTION REACTIONS

Oxidation Reactions in Everyday Life			
		Name _____	
Burning of iron is an example of a redox reaction.	TRUE	FALSE	Justify:
Combustion reactions only occur in the presence of oxygen.	TRUE	FALSE	Justify:
Corrosion is the process of metal gaining electrons.	TRUE	FALSE	Justify:
Rusting of iron can be accelerated by exposure to saltwater.	TRUE	FALSE	Justify:
Combustion reactions release energy in the form of heat and light.	TRUE		
Corrosion is a slow process that occurs in metals.	TRUE		
Combustion reactions involve the reduction of oxygen.	TRUE		
The rusting of iron is an example of a non-redox reaction.	TRUE		

Steps for Balancing Redox Equations	
<p>Balance the following redox equation using the acidic medium method:</p> $\text{Cr}_2\text{O}_7^{2-} + \text{Fe}^{2+} \rightarrow \text{Cr}^{3+} + \text{Fe}^{3+}$	<p>Balance the following redox equation using the basic medium method:</p> $\text{MnO}_4^- + \text{C}_2\text{O}_4^{2-} \rightarrow$
<p>Balance the following redox equation using the half-reaction method:</p> $\text{Cr}_2\text{O}_7^{2-} + \text{Fe}^{2+} \rightarrow \text{Cr}^{3+} + \text{Fe}^{3+}$	<p>Balance the following redox equation using the half-reaction method:</p> $\text{Cr}_2\text{O}_7^{2-} + \text{Fe}^{2+} \rightarrow \text{Cr}^{3+} + \text{Fe}^{3+}$



Oxidation And Reduction Packet

Katrin Zwirgmaier



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updated with newer clearer or better images As a new feature each chapter now contains intriguing openended questions highlighting What We Don t Know introducing students to challenging areas of future research Updated end of chapter problems reflect new research discussed in the text and these problems have been expanded to all chapters by adding questions on developmental biology tissues and stem cells pathogens and the immune system **Burton's Microbiology**

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