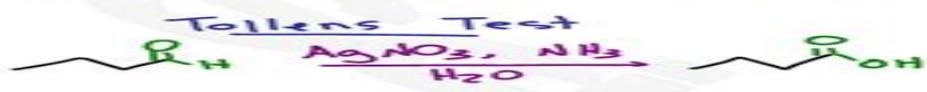
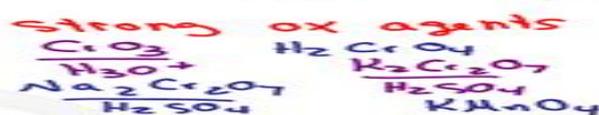


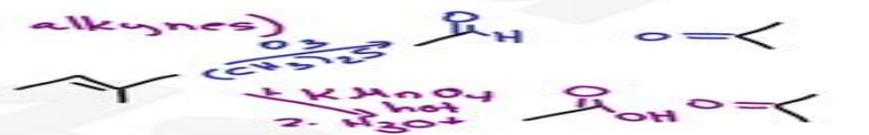
# OXIDATION & REDUCTION

Redox Video Series - [leah4sci.com/Redox](http://leah4sci.com/Redox)

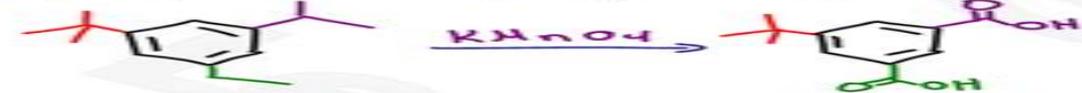
Oxidation = Gain O bonds, Lose H bonds



Alkene Oxidation (also alkynes)



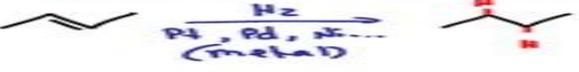
Side chain Oxidation



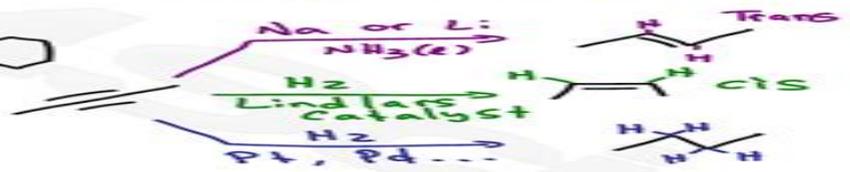
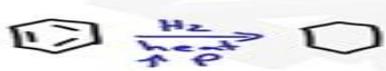
only 1° 2° + 3° get oxidized  
4° = N/R

Reduction = Gain H bonds, Lose O bonds

Alkene Reduction



Alkyne Reduction



Birch Reduction



Reduction With Metal Hydrides



LiAlH<sub>4</sub> = strong



NaBH<sub>4</sub> = weak



Clemmenson



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# Oxidation And Reduction Study Guide

**Jennie Victoria Davenport ,Judith Bess  
Perkins**



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## **Table of Contents Oxidation And Reduction Study Guide**

1. Understanding the eBook Oxidation And Reduction Study Guide
  - The Rise of Digital Reading Oxidation And Reduction Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Oxidation And Reduction Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Oxidation And Reduction Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Oxidation And Reduction Study Guide
  - Personalized Recommendations
  - Oxidation And Reduction Study Guide User Reviews and Ratings
  - Oxidation And Reduction Study Guide and Bestseller Lists
5. Accessing Oxidation And Reduction Study Guide Free and Paid eBooks
  - Oxidation And Reduction Study Guide Public Domain eBooks
  - Oxidation And Reduction Study Guide eBook Subscription Services
  - Oxidation And Reduction Study Guide Budget-Friendly Options

- 
6. Navigating Oxidation And Reduction Study Guide eBook Formats
    - ePub, PDF, MOBI, and More
    - Oxidation And Reduction Study Guide Compatibility with Devices
    - Oxidation And Reduction Study Guide Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Oxidation And Reduction Study Guide
    - Highlighting and Note-Taking Oxidation And Reduction Study Guide
    - Interactive Elements Oxidation And Reduction Study Guide
  8. Staying Engaged with Oxidation And Reduction Study Guide
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Oxidation And Reduction Study Guide
  9. Balancing eBooks and Physical Books Oxidation And Reduction Study Guide
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Oxidation And Reduction Study Guide
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Oxidation And Reduction Study Guide
    - Setting Reading Goals Oxidation And Reduction Study Guide
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Oxidation And Reduction Study Guide
    - Fact-Checking eBook Content of Oxidation And Reduction Study Guide
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements

- 
- Interactive and Gamified eBooks

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