

Assessment

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Understanding Concepts

- What is a stable electron configuration?
- What does each dot in an electron dot diagram represent?
- What process changes atoms into ions?
- What keeps the ions in their fixed positions within a crystal lattice?
- What are subscripts used for in chemical formulas?
- Explain why a melted ionic compound is a good conductor of electric current, but a solid ionic compound is a poor conductor of electric current.
- What distinguishes single, double, and triple covalent bonds?
- Explain why the covalent bonds in molecules of O_2 and N_2 are always nonpolar.
- Explain why, in a covalent bond between oxygen and hydrogen, the hydrogen atom has a partial positive charge and the oxygen atom has a partial negative charge.
- What is the name of the binary compound formed from potassium and iodine?
- Write the formulas for the compounds called copper(I) chloride and copper(II) chloride.
- Name the compounds represented by the space-filling models labeled A, B, and C.



- In general, what determines the strength of metallic bonds?
- What properties of copper and tin change when these metals are mixed together to form bronze?
- What advantage of magnesium is retained in magnesium alloys? What disadvantage is reduced?

Chemical Bonds 187

Assessment



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Reviewing Content

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| 7. d | 8. b | 9. c |
| 10. a | | |

Understanding Concepts

- In a stable electron configuration, the highest occupied energy level is filled with electrons.
- Each dot represents a valence electron.
- The transfer of electrons.
- Attractions between neighboring cations and anions keep the ions in fixed positions within the lattice.
- A subscript is used to show the number of atoms of an element in a molecule or the ratio of ions in a crystal lattice.
- When an ionic compound melts, ions can move away from their fixed locations in the crystal lattice.
- Two atoms share two electrons in a single bond, four in a double bond, and six in a triple bond.
- The covalent bonds in molecules of elements are always nonpolar because the atoms have the same attraction for electrons.
- The oxygen atom has a greater attraction for electrons than the hydrogen atom does.
- Potassium iodide
- CuCl and CuCl_2
- A is sulfur trioxide, B is carbon monoxide, and C is nitrogen dioxide
- In general, the more valence electrons a metal has, the stronger the metal bonds are.
- A mixture of copper and tin is harder and stronger than either metal in its pure form.
- The advantage that is retained is that magnesium is a lightweight metal. The disadvantage that is reduced is that magnesium is a soft metal.

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Pearson Physical Science Chapter 4 Review Answers

Kathleen Armour



Pearson Physical Science Chapter 4 Review Answers:

Scott Foresman Science, 2008 *The Essential Peirce, Volume 2* Charles Sanders Peirce, 1992 A convenient two volume reader's edition makes accessible to students and scholars the most important philosophical papers of the brilliant American thinker Charles Sanders Peirce Back cover Conceptual Physical Science Paul G. Hewitt, John Suchocki, Leslie A. Hewitt, 2004 Conceptual Physical Science Third Edition takes learning physical science to a new level by combining Hewitt's leading conceptual approach and friendly writing style in a new edition that provides stronger integration of the sciences more quantitative coverage and a wealth of new media resources to help professors in class and students out of class The book's consistent high quality coverage includes five new chapters on chemistry astronomy and earth science for an even more balanced approach to physical science New Looking Forward and Looking Back boxes connect themes and concepts throughout the book helping students see the big picture More computational coverage eg Figuring Physical Science in chapter calculation allows students to practice the quantitative skills they need to master the concepts of physical science and be able to apply their knowledge Looking Forward and Looking Back boxes in every chapter connect themes and concepts throughout the book helping students see the big picture of physical science Powerful media package includes a comprehensive suite of award winning interactive online tutorials that offer students 24/7 help A media gri **The Essential Peirce, Volume 2 (1893-1913)** The Peirce Edition Project, 1998-06-22 Praise for Volume 1 a first rate edition which supersedes all other portable Peirces all the Peirce most people will ever need Louis Menand The New York Review of Books Volume 2 of this convenient two volume chronological reader's edition provides the first comprehensive anthology of the brilliant American thinker Charles Sanders Peirce's mature philosophy A central focus of Volume 2 is Peirce's evolving theory of signs and its application to his pragmatism **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Physical Science Teacher's Desk Reference Prentice-Hall Staff, 1993 **Athenaeum and Literary Chronicle**, 1849 *Preparing for The Occupational Therapy Assistant National Board Exam: 45 Days and Counting* Rosanne DiZazzo-Miller, Fredrick D. Pociask, 2017-08-30 Preparing for the Occupational Therapy Assistant Board Exam 45 Days and Counting provides a comprehensive overview for occupational therapy assistant students preparing to take the Certified Occupational Therapy Assistant COTA exam Each new print copy includes Navigate 2 Preferred Access that unlocks a complete eBook Study Center Homework and Assessment Center Navigate 2 TestPrep with over 500 practice questions **Resources in education**, 1987-03 273 technical questions and answers for job interview Offshore Drilling Rigs Petrogav International Oil & Gas

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Joubert, Ansie E. Kitching, Michael L. Marumo, Zahraa McDonald, Nicholas T. Mollo, Louis J. Oosthuizen, Ferdinand J. Potgieter, Bianke van Rooyen, Marius H. Smit, Amos T. Thebenyane, Andre van der Bijl, Johannes (Hannes) L. van der Walt, 2019-12-12 This book is a collection of chapters based on original research dealing with issues of discipline and disciplinary practices in educational institutions The aim of the book is to provide a scholarly and scientific perspective on the current state of discipline and disciplinary practices in schools and tertiary education settings The issue of discipline is investigated from diverse paradigmatic and methodological perspectives presenting empirical as well as also philosophical research The empirical perspective includes quantitative positivistic qualitative interpretive and mixed methods pragmatic designs and worldviews This book offers a ground breaking contribution to the field of learner and student discipline with insights into disciplinary practices and issues in educational institutions not hitherto researched such as Technical Vocational Education and Training colleges and universities

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