

All Persons

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# DEVEL PROGRESS

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### Radioactive Dating Game (22 Points) ~ Answer Key

Go To: <http://phet.colorado.edu/en/simulation/radioactive-dating-game>

**Purpose:** You will use the radioactive decay rate and original-daughter element ratios of carbon-14 and uranium-238 to determine the ages of different objects.

**Procedure:**

1. Load PhET Radioactive Dating Game
2. Click on tab for Decay Rates
3. (1pt) Select Carbon-14. Using the graph, the estimated half-life for C-14 is \_\_\_\_\_ years.
4. Move the bucket later all the way to the right. This will place 1000 C-14 atoms onto the screen.



- a. (1pt) Click on the Start/Stop to stop the C-14 decay as you get close to one half-life. Use the Step button to advance the decay to one half-life.
  - After 1 half-life, how many C-14 atoms of the 1000 original remain? Use the simulation, also? Just write about half of the original). \_\_\_\_\_
- b. (1pt) Use the Start/Stop and Step buttons to reach two half-lives. After two half-lives, how many undecayed C-14 atoms remain?
  - Calculate the percent of original undecayed C-14 atoms present after 2 half-lives? \_\_\_\_\_
- c. (1pt) Use the Start/Stop and Step buttons to reach three half-lives. After three half-lives, how many undecayed C-14 atoms remain?
  - Calculate the percent of original undecayed C-14 atoms present after 3 half-lives? \_\_\_\_\_
- d. (2pts) Repeat Steps (a) to (c) with uranium-238.
  - Estimated half-life for U-238 is \_\_\_\_\_ years.
  - After 1 half-life, how many U-238 atoms of the 1000 original remain? \_\_\_\_\_
  - Calculate the percent of original undecayed C-14 atoms present after 2 half-lives? \_\_\_\_\_
  - Calculate the percent of original undecayed C-14 atoms present after 3 half-lives? \_\_\_\_\_
- e. (1pt) Based on the results of (a) to (d), explain the meaning of the word "decay" in one sentence.
   
\_\_\_\_\_

# Pearson Radioactive Dating Of Rocks Answer Key

**Padhraic Smyth**

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Hydrosphere V.I. Ferronsky,V.A. Polyakov,2012-03-06 This book covers the distribution hydrochemistry and geophysics of the naturally occurring stable isotopes namely hydrogen oxygen and radioactive tritium carbon and other cosmogenic and radiogenic isotopes of the uranium thorium series in the oceans and in atmosphere the earth s surface and ground water The use of environmental isotopes in the three main areas of natural waters is discussed origin dynamics and residence time in natural reservoirs The origin of the hydrosphere is examined in the light of isotopic new cosmochemical and recent theoretical results The book will be of interest to scientists and researchers who use environmental isotopes in solving scientific and practical problems in hydrology hydrogeology oceanography meteorology hydrogeochemistry and cosmochemistry Lecturers students and postgraduates in these fields will also find it useful    **Bibliography of North**

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