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web May 28 2020 write the balanced equation for the reaction and calculate how much  $\text{SiO}_2$  is required to produce 3.00 kg of  $\text{SiC}$  answer  $\text{Ce} + \text{SiO}_2 \rightarrow 3 \text{C} + \text{SiC} + 2 \text{CO}$

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equations  $\text{N}_2$   $\text{F}_2$   $\text{NF}$   $\text{C}_6\text{H}_{10}$   $\text{O}_2$   $\text{CO}$   $\text{H}_2$   $\text{HBr}$   $\text{KHCO}_3$   $\text{H}_2\text{O}$   $\text{KBr}$   $\text{Na}_2\text{SO}_3$   $\text{Ga}_2\text{SO}_3$   $\text{SNO}$   $\text{NF}_3$   $\text{N}_2\text{O}$   $\text{CO}_2$   $\text{NaBr}$  solve the following stoichiometry grams grams problems 6

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web gabrielle m 9 years ago i m working on chemical reactions double and single replacement on flvs now my first question for this video is how do you have 4 aluminum atoms when it says  $2\text{Al}$  subscript 2 do you just add the 2 and the subscript 2 same thing goes with the  $\text{O}_3$  i am so lost

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web google classroom balance the following chemical equation  $\text{Mg} + \text{OH}^- \rightarrow \text{Mg}^{2+} + \text{OH}^-$   $\text{HCl} \rightarrow \text{HCl}$   $\text{MgCl}_2$   $\text{H}_2\text{O}$  note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem 7 4 1 x x y y theta  $\theta$  pi  $\pi$  8 5 2 0 9 6

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