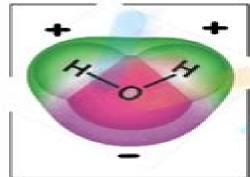
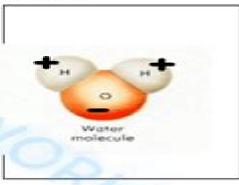
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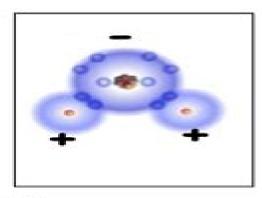
Name______ Class_____

Water is everywhere. It's in the air we breathe. It's in our sink faucets. It's in every cell of our body. Water is an unusual substance with **many special properties**.

INTRODUCTION: The Water Molecule







What do you observe about the water molecule in these models?

- 1. _____
- 2. _____
- 3

READ: Water is a polar molecule. That is, while a water molecule is neutral as a whole, it has a slightly positive charge on the oxygen side and a slightly negative charge on the hydrogen side. Water is found on Earth as a solid, liquid, and gas. This is because Earth is a very special planet with just the right range of temperatures and air pressures. Earth is said to be at the triple point for water.

QUESTIONS:

- How many atoms are in a water molecule? Be specific.
- 2. Why does a water molecule have polarity?
- 3. Why is Earth said to be at the triple point for water?

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