

# Answer Key

## Unit 1 Biochemistry

### Answers to Unit 1 Preparation Questions Assessing Student Readiness (Student textbook pages 4–7)

- Sketches should depict a central atomic nucleus surrounded by four valence electrons (or six electrons in total), or six neutrons and six protons within the nucleus.
- b
- a
- metal, non-metal ion
- Since the atoms in a glucose molecule share valence electrons, the bonds between these atoms must be covalent.
  - Glucose is a stable molecule because its covalent bonds are stable and strong.
- water—molecular, atoms (both non-metals) are joined by covalent bonds
  - carbon dioxide—molecular, atoms (both non-metals) are joined by covalent bonds
  - glucose—molecular, atoms (two non-metals and uncharged hydrogen) are joined by covalent bonds
  - sodium chloride—ionic, ions (a metal and a non-metal) joined by an ionic bond
  - calcium phosphate—ionic, ions (a metal and a polyatomic ion) joined by ionic bonds
  - methane—molecular, atoms (a non-metal and uncharged hydrogen) joined by covalent bonds
  - oxygen—molecule or molecular element, atoms joined by double covalent bond, but not a molecular compound because it is formed from only one element (two oxygen atoms)
  - ammonia—molecular, atoms (a non-metal and uncharged hydrogen) joined by covalent bonds
- chloride
  - sulphate/sulfate
  - magnesium ion
  - copper (II) ion
- $\text{OH}^-$
  - $\text{S}^{2-}$
  - $\text{Al}^{3+}$
  - $\text{Fe}^{3+}$
- $\text{H}_2\text{O}$  or water
- $\text{water} \rightarrow \text{hydrogen} + \text{oxygen}$
- d
- sodium hydroxide + hydrochloric acid  $\rightarrow$   
water + sodium chloride
- Since the pH of the solution increased, the student must have added a base to the solution.
  - One of: When the second substance was added, it released hydroxide ions ( $\text{OH}^-$ ), which combined with protons ( $\text{H}^+$ ) to produce water. The second substance accepted protons ( $\text{H}^+$ ) from the solution.
- decomposition
  - neutralization
  - decomposition
  - synthesis
  - neutralization
  - synthesis
  - decomposition
  - synthesis
  - synthesis
  - synthesis
- pH 7
- c
- One of: enzymes, cell structure, membrane transport, muscle
- A molecule is a substance composed of two or more atoms that are bonded together covalently. A macromolecule is a large, complex molecule. Usually, the subunits of a macromolecule are small molecules bonded together.
- An enzyme is a macromolecule that speeds up chemical reactions.
- c
- c
- b

# Review Unit 1 Biochemistry Answers

**Rachel Sandford**



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