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Answer & Solutions for NEET 2022_(Re-Exam)

Chemistry

51. Match List - I with List - II :

List - I (quantum number)	List - II (orbital)
(a) $n = 2, \ell = 1$	(i) 2s
(b) $n = 3, \ell = 2$	(ii) 3s
(c) $n = 3, \ell = 0$	(iii) 2p
(d) $n = 2, \ell = 0$	(iv) 3d

Choose the correct answer from the options given below :

- (1) (a) - (iii), (b) - (iv), (c) - (ii), (d) - (i)
 (2) (a) - (iii), (b) - (iv), (c) - (i), (d) - (ii)
 (3) (a) - (iv), (b) - (iii), (c) - (i), (d) - (ii)
 (4) (a) - (iv), (b) - (iii), (c) - (ii), (d) - (i)

Sol. Answer (1)

$\ell = 0 \Rightarrow$ s-subshell

$\ell = 1 \Rightarrow$ p-subshell

$\ell = 2 \Rightarrow$ d-subshell

$\ell = 3 \Rightarrow$ f-subshell

$\therefore n = 2, \ell = 1 \Rightarrow 2p$

$n = 3, \ell = 2 \Rightarrow 3d$

$n = 3, \ell = 0 \Rightarrow 3s$

$n = 2, \ell = 0 \Rightarrow 2s$

52. The density of the solution is 2.15 g mL^{-1} , then mass of 2.5 mL solution in correct significant figures is :

- (1) 53.75 g (2) $5375 \times 10^{-3} \text{ g}$
 (3) 5.4 g (4) 5.38 g

Sol. Answer (3)

In case of multiplication and division, the final result should be reported as having the same number of significant digits as the number with least number of significant digits.

$$\therefore \text{Density} = \frac{\text{Mass}}{\text{Volume}}$$

$$\begin{aligned} \text{So, Mass} &= 2.15 \times 2.5 \\ &= 5.375 \\ &\approx 5.4 \text{ g} \end{aligned}$$

53. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

Assertion (A) :

Chlorine is an electron withdrawing group but it is ortho, para directing in electrophilic aromatic substitution.

Reason (R) :

Inductive effect of chlorine destabilises the intermediate carbocation formed during the electrophilic substitution, however due to the more pronounced resonance effect, the halogen stabilises the carbocation at ortho and para positions.

In the light of the above statements, choose the most appropriate answer from the options given below :

- (1) (A) is not correct but (R) is correct.
 (2) Both (A) and (R) are correct and (R) is the correct explanation of (A).
 (3) Both (A) and (R) are correct but (R) is not the correct explanation of (A).
 (4) (A) is correct but (R) is not correct.

Neet Exam Paper 2answer Key

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Table of Contents Neet Exam Paper 2answer Key

1. Understanding the eBook Neet Exam Paper 2answer Key
 - The Rise of Digital Reading Neet Exam Paper 2answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Neet Exam Paper 2answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Neet Exam Paper 2answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Neet Exam Paper 2answer Key
 - Personalized Recommendations
 - Neet Exam Paper 2answer Key User Reviews and Ratings
 - Neet Exam Paper 2answer Key and Bestseller Lists

5. Accessing Neet Exam Paper 2answer Key Free and Paid eBooks
 - Neet Exam Paper 2answer Key Public Domain eBooks
 - Neet Exam Paper 2answer Key eBook Subscription Services
 - Neet Exam Paper 2answer Key Budget-Friendly Options
6. Navigating Neet Exam Paper 2answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Neet Exam Paper 2answer Key Compatibility with Devices
 - Neet Exam Paper 2answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Neet Exam Paper 2answer Key
 - Highlighting and Note-Taking Neet Exam Paper 2answer Key
 - Interactive Elements Neet Exam Paper 2answer Key
8. Staying Engaged with Neet Exam Paper 2answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Neet Exam Paper 2answer Key
9. Balancing eBooks and Physical Books Neet Exam Paper 2answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Neet Exam Paper 2answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Neet Exam Paper 2answer Key
 - Setting Reading Goals Neet Exam Paper 2answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Neet Exam Paper 2answer Key
 - Fact-Checking eBook Content of Neet Exam Paper 2answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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