

PHYSICAL AND CHEMICAL PROPERTIES AND CHANGES

Name _____ Key _____

PHYSICAL PROPERTY

1. observed with senses
2. determined without destroying matter

CHEMICAL PROPERTY

1. indicates how a substance reacts with something else
2. matter will be changed into a new substance after the reaction

Identify the following as a chemical (C) or physical property (P):

- ___P___ 1. blue color
- ___P___ 2. density
- ___C___ 3. flammability (burns)
- ___C___ 4. solubility (dissolves)
- ___C___ 5. reacts with acid
- ___C___ 6. supports combustion
- ___P___ 7. sour taste

- ___C___ 8. melting point
- ___C___ 9. reacts with water
- ___P___ 10. hardness
- ___C___ 11. boiling point
- ___P___ 12. luster
- ___P___ 13. odor
- ___C___ 14. reacts with air

PHYSICAL CHANGE

1. a change in size, shape, or state
2. no new substance is formed

CHEMICAL CHANGE

1. a change in the physical and chemical properties
2. a new substance is formed

Identify the following as physical (P) or chemical (C) changes.

- ___C___ 1. NaCl (Table Salt) dissolves in water.
- ___C___ 2. Ag (Silver) tarnishes.
- ___P___ 3. An apple is cut.
- ___P___ 4. Heat changes H₂O to steam.
- ___C___ 5. Baking soda reacts to vinegar.
- ___C___ 6. Fe (Iron) rusts.
- ___P___ 7. Alcohol evaporates.
- ___P___ 8. Ice melts.

- ___C___ 9. Milk sours.
- ___C___ 10. Sugar dissolves in water.
- ___C___ 11. Wood rots.
- ___C___ 12. Pancakes cook.
- ___P___ 13. Grass grows.
- ___P___ 14. A tire is inflated.
- ___C___ 15. Food is digested.
- ___P___ 16. Paper towel absorbs water.

Physical and Chemical Changes

Part A

Can you recognize the chemical and physical changes that happen all around us? If you change the way something looks, but haven't made a new substance, a **physical change (P)** has occurred. If the substance has been changed into another substance, a **chemical change (C)** has occurred.

1.	P	An ice cube is placed in the sun. Later there is a puddle of water. Later still the puddle is gone.
2.	C	Two chemicals are mixed together and a gas is produced.
3.	C	A bicycle changes color as it rusts.
4.	P	A solid is crushed to a powder.
5.	C	Two substances are mixed and light is produced.
6.	C	A piece of ice melts and reacts with sodium.
7.	P	Mixing salt and pepper.
8.	P	Chocolate syrup is dissolved in milk.
9.	C	A marshmallow is toasted over a campfire.
10.	P	A marshmallow is cut in half.

Physical And Chemical Properties Pogil Answers

Carl L. Yaws



Physical And Chemical Properties Pogil Answers:

Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes **Chemistry Education** Javier García-Martínez, Elena Serrano-Torregrosa, 2015-05-04 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry students **Properties of Matter Gr. 5-8** George Graybill, 2007-09-01 Discover what matter is and what it isn t Our resource breaks down the physical and chemical properties of matter to make it more accessible to students Start off by identifying matter as atoms particles and molecules Then explore the three states of matter solid liquid and gas Determine whether something is transparent opaque or translucent List three physical changes and three chemical changes that could happen in the kitchen Conduct an experiment to see chemical change in action Describe the steps necessary when separating a mixture Experiment with photosynthesis an important chemical change Aligned to the Next Generation Science Standards and written to Bloom s Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included *Choice* , 2009 [Physical and Chemical Changes](#) Edward P. Ortleb, Richard Cadice, 1993-09-01 General chemistry information including everything from matter to radioactivity For grades 5 to 9 **Properties of Matter: Physical Changes vs. Chemical Changes Gr. 5-8** George Graybill, 2015-09-01 This is the chapter slice Physical Changes vs Chemical Changes from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and

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Physical and Chemical Properties and Changes Jenny Karpelenia,2014-06-23 Physical and chemical properties and changes can be used to describe and classify all matter

Chemical Changes Rebecca L. Johnson,2008-06-01 Everything around us is matter Physical properties describe what matter is like and chemical properties describe what matter can do In a physical change the physical properties of matter change from one state to another During a chemical change the chemical properties change and produce a new substance Signs of these changes include sound light color smell or bubbles

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Accidental Explosions Louis A. Medard,Peter Fawcett,1990-01-01

Vol. 1 Louis A. Medard,1989

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Physical Properties of Chemical Compounds Robert Rickert Dreisbach,1955

Physical Properties of Chemical Substances Dow Chemical Company,1952

States of Matter Delores Boufford,1999-09-01 The activities in this book explain elementary concepts in the study of chemistry including the physical properties of matter solids liquids and gases surface tension capillary action and expansion General background information suggested activities questions for discussion and answers are included

Chemical Properties Handbook Carl L. Yaws,2001

Physical Properties of Chemical Compounds ,1955

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