

Name \_\_\_\_\_ Date \_\_\_\_\_

# PHYSICAL & CHEMICAL CHANGE

In a physical change, the original substance still exists; it has only changed in form (i.e. size, shape, state of matter, etc). In addition, a physical change is always reversible - you can get back what you started with! In a chemical change, however, a completely new substance is produced - you can never get back what you started with! Energy (heat) changes typically accompany chemical changes, but can sometimes occur in physical changes as well. Do not assume a chemical change is taking place just because energy is involved! Think: Do I have a totally different substance than I started with? If so, this is always a chemical change.

Read the following statements. Classify the following as being a physical or chemical change.

PHYSICAL CHANGE	CHEMICAL CHANGE	STATEMENTS
		An ice cube melts on a hot summer day.
		Green grass making its own food.
		Hydrochloric acid reacts with potassium hydroxide to produce salt, water, and heat.
		Pancakes cook on a griddle.
		non-metals do not conduct heat or electricity well
		On a cold fall morning, water vapor in the atmosphere condenses into dew on the grass.
		A tire is deflated (the air is let out).
		Potassium chlorate decomposes to potassium chloride and oxygen gas.
		When placed in water, a sodium pellet catches on fire as hydrogen gas is given off and sodium hydroxide forms.
		Wood rots out in the rain.
		An iron nail rusts when left out in the rain.
		A pellet of sodium is sliced in two.
		Food is digested in the stomach
		Milk sours after being kept past the expiration date.
		A pellet of sodium is sliced in two.
		Sugar dissolves in water.
		Sodium hydroxide dissolves in water.

# Physical Change Vs Chemical Change Worksheet

## Chemistry If8766

**AW Rasmussen**



## **Physical Change Vs Chemical Change Worksheet Chemistry If8766:**

Physical and Chemical Changes (eBook) Edward P. Ortleb, Richard Cadice, 1993-09-01 This book presents a program of basic studies in physical and chemical changes of matter The definition of matter is presented along with explanations of states and properties of matter Topics include atoms molecules elements compounds mixtures solutions symbols and formulas Each of the twelve teaching units in this book is introduced by a color transparency print books or PowerPoint slide eBooks that emphasizes the basic concept of the unit and presents questions for discussion Reproducible student pages provide reinforcement and follow up activities The teaching guide offers descriptions of the basic concepts to be presented background information suggestions for enrichment activities and a complete answer key **Properties of Matter:**

**Physical Changes vs. Chemical Changes Gr. 5-8** George Graybill, 2015-09-01 This is the chapter slice Physical Changes vs Chemical Changes from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

Physical vs. Chemical Changes in Matter | Understanding Chemical Properties of Matter | Grade 6-8 Physical Science Baby Professor, 2024-04-15 Dive into the transformative world of matter with Physical vs Chemical Changes in Matter This essential guide for grades 6 8 demystifies the complexities behind physical and chemical changes It lays a solid foundation for understanding chemical properties from melting ice to rusting metal A must have for educators homeschooling parents and school libraries it s aligned with the US STEM curriculum ensuring young scientists grasp these fundamental concepts Ignite curiosity about the world s material makeup today **Changing Matter** Tracy Nelson

Maurer, 2014-05-30 This title teaches students that everything is made of matter and that physical changes create different forms or states of matter Examples of these different states are presented in easy to understand text The book also introduces students to the law of conservation of mass *Chemical Changes* Rebecca L. Johnson, 2008-06-01 Everything around us is matter Physical properties describe what matter is like and chemical properties describe what matter can do In a physical change the physical properties of matter change from one state to another During a chemical change the chemical properties change and produce a new substance Signs of these changes include sound light color smell or bubbles

Physical and Chemical Changes in Matter : Chemistry Science Book for Grade 2 | Children's Chemistry Books Baby Professor, 2022-12-01 You are now ready to learn about the physical and chemical changes in matter which is an introduction

to chemistry The first chapter teaches you about elements or the basic blocks of matter Next you will read about what is physical change and its types The third chapter discusses what is chemical change and its examples Start reading today

*Chemical Changes* Rebecca L. Johnson,2008-06-01 Matter has both physical and chemical properties During a chemical change the chemical properties of a substance change and one or more new substances form Every chemical reaction involves a transfer of energy During a chemical reaction atoms in matter are rearranged but no atoms are gained or lost

*Chemical Changes* Lynnette Brent,2008-08-15 An introduction to how chemicals react and change     *Physical Vs. Chemical Changes in Matter Understanding Chemical Properties of Matter Grade 6-8 Physical Science* Baby Professor,2024-01-04 Dive into the transformative world of matter with Physical vs Chemical Changes in Matter This essential guide for grades 6 8 demystifies the complexities behind physical and chemical changes It lays a solid foundation for understanding chemical properties from melting ice to rusting metal A must have for educators homeschooling parents and school libraries it s aligned with the US STEM curriculum ensuring young scientists grasp these fundamental concepts Ignite curiosity about the world s material makeup today     *Chemical Changes* Rebecca Kraft Rector,2019-07-15 A burning log a rotting banana and a rusting bicycle These are examples of chemical changes With this accessible narrative readers are introduced to the basics of chemical changes Easy to follow explanations cover matter forms properties and changes Further detail introduces how to make chemical changes and how to recognize them Color photographs fast facts and an engaging activity expand and reinforce the newly learned ideas This book supports Next Generation Science Standards correlating directly with the NGSS Grade 2 standard to classify different kinds of materials by their observable properties     **Physical and Chemical Change** Alan V. Jones,Nigel Heslop,Mike Clemmet,1995     *Understanding Physical and Chemical Changes* Richard G. Cote,Darcy O. Blauvelt,2012-06 In *Understanding Physical and Chemical Changes* a science unit students work to understand the concepts and definitions of physical and chemical change Grades 6 8     [Bad Reaction! Evidence of a Chemical Reaction | Endothermic vs Exothermic Reactions | Grade 6-8 Physical Science](#) Baby Professor,2024-04-15 Unlock the mysteries of chemical reactions with this engaging book tailored for middle schoolers Perfect for educators homeschooling parents and librarians it delves into the transformative world of endothermic and exothermic reactions showcasing the dynamic nature of chemistry Discover the signs and wonders of chemical changes from fireworks to baking pies emphasizing these reactions critical role in the US STEM curriculum Equip your students with the knowledge to explore the fascinating changes around us     [Chemical Change](#) Darlene R. Stille,2005-09 Learn about chemical changes and what causes them     **Chemical Changes** ,2003-01-14 Any change that results in the formation of a new kind of matter is a chemical change Energy is always involved in a chemical change sometimes with spectacular results     *Science Workshop Series: Chemical changes* Seymour Rosen,2000 This program presents science concepts in areas of biology earth science chemistry and physical science in a logical easy to follow design that challenges without overwhelming This flexible program

consists of 12 student texts that can easily supplement an existing science curriculum or be used as a stand alone course  
Reading Level 4 5 Interest Level 6 12      **Chemical Reactions 6-Pack** Jenna Winterberg,2015-09-20 Learn about six types of chemical reactions activation energy and hopping electrons reactivity catalysts and inhibitors physical changes of mixtures and more with this high interest nonfiction title This 6 Pack provides five days of standards based activities that will engage fifth grade students support STEM education and build content area literacy in life science It includes vibrant images fun facts helpful diagrams and text features such as a glossary and index The hands on Think Like a Scientist lab activity aligns with Next Generation Science Standards NGSS The accompanying 5E lesson plan incorporates writing to increase overall comprehension and concept development and features Step by step instructions with before during and after reading strategies Introductory activities to develop academic vocabulary Learning objectives materials lists and answer key Science safety contract for students and parents      *Chemical Changes* ,2004 Any change that results in the formation of a new kind of matter is a chemical change Energy is always involved in a chemical change sometimes with spectacular results

**Exploring Matter & Physical Changes** Jessica Rusick,2022-08-01 This title provides an overview of matter and physical changes Text includes a simple overview of matter and examines properties states phases and atoms Atomic theory is introduced Information is explained using real world examples and supported with graphics and photos This book concludes with two simple kid friendly experiments Aligned to Common Core standards and correlated to state standards Checkerboard Library is an imprint of Abdo Publishing a division of ABDO      **Chemical Changes** Rebecca L. Johnson,2008-06-01 Physical and chemical properties of matter can change In a chemical reaction reactants change into products Energy is either absorbed or released but matter is always conserved Bonds in reactants are broken and atoms are rearranged to form products The pH scale measures the strength of acids and bases found in many substances

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web answer 2 56 m s this is an example of energy transformation from potential energy at the highest point the point of release to kinetic energy at the lowest position since gravity is the only force doing work tension acts perpendicular to the displacement so it does not do work the total mechanical energy is conserved

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web answer adghk a true momentum is a vector quantity like all vector quantities the momentum of an object is not fully

described until the direction of the momentum is identified momentum like other vector quantities is subject to the rules of vector operations b false the joule is the unit of work and energy the kg m s is the

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web the type of the charge is dependent upon whether there are more protons or more electrons present in the object if there are more protons then the charge is otherwise assign it a value useful web links neutral vs charged objects 33 34 35 36 37 38 39 40 41 42 43 34

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web mathematically it is computed using the following equation power work time or  $p = \frac{w}{t}$  the standard metric unit of power is the watt as is implied by the equation for power a unit of power is equivalent to a unit of work divided by a unit of time thus a watt is equivalent to a joule second



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web power an object can store energy as the result of its position for example the heavy ball of a demolition machine is storing energy when it is held at an elevated position this stored energy of position is referred to as potential energy similarly a drawn bow is able to store energy as the result of its position

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web answer a for any given situation the work done by a force can be calculated using the equation  $w = f d \cos \theta$  where  $f$  is the force doing the work  $d$  is the displacement of the object and  $\theta$  is the angle between the force and the displacement

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web show answer problem 3 lamar gant u s powerlifting star became the first man to deadlift five times his own body weight in 1985 deadlifting involves raising a loaded barbell from the floor to a position above the head with outstretched arms determine the work done by lamar in deadlifting 300 kg to a height of 0.90 m above the ground

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web energy bar charts or even work energy bar charts are conceptual tools that help to illustrate how the energy changes both the form and the amount as the object moves from the initial state to the final state

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web check your understanding use your understanding of the work energy theorem to answer the following questions then click the button to view the answers 1 consider the falling and rolling motion of the ball in the following two resistance free situations