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Fission and Fusion

11. The number and mass of nuclei changes in a fission process. The number _____ and mass _____ change.
12. How is the energy released? How about is the difference between the mass of the separate particles before and the products in a fission reaction?
13. Calculate the energy released when a nucleus of an isotope of Uranium splits into two nuclei and some neutrons in the process. Find out the splitting of one nucleus and its mass in several grams (grams).
14. Calculate the energy released when the entire mass of a 1000 kg of Uranium splits into two nuclei and some neutrons in the process and its mass in 1000 kg of Uranium.

$$E = mc^2 = \frac{1000 \text{ kg}}{1000} \times (3 \times 10^8 \text{ m/s})^2 = 9 \times 10^{13} \text{ J}$$

$$E = mc^2 = (1000 \text{ kg} \times 1000 \text{ kg} \times 1000 \text{ kg}) \times (3 \times 10^8 \text{ m/s})^2 = 9 \times 10^{16} \text{ J}$$

15. Balance and find the following nuclear reactions. Calculate the energy released in fission.



16. a. Calculate the following is an example of a nuclear reaction?

(a) The nuclear reaction

(b) The nuclear reaction

(c) The nuclear reaction

(d) The nuclear reaction

17. a. The mass of the reaction and mass in a given process is controlled by the law of

(a) conservation of mass

(b) conservation of mass

(c) conservation of mass

(d) conservation of mass

18. Calculate the energy released in the following nuclear reaction.

Reaction: $\text{Uranium} \rightarrow \text{Ba} + \text{Kr} + \text{Neutrons}$ **Energy:** $9 \times 10^{13} \text{ J}$ **Mass:** 1000 kg **Energy:** $9 \times 10^{13} \text{ J}$ **Mass:** 1000 kg

Phys Sci Ch 14 Fission And Fusion Worksheet Answers

Rachel S Tattersall



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