Balancing Chemical Equations Worksheet

1.
$$2 H_2 + 0_2 \rightarrow 2 H_2O$$

3.
$$S_8 + 12 O_2 \rightarrow 8 SO_3$$

4.
$$2 N_2 + 0_2 \rightarrow 2 N_20$$

5.
$$2 \text{ HgO} \rightarrow 2 \text{ Hg} + 0$$

6.
$$6 \, CO_2 + 6 \, H_2O \Rightarrow C_6H_{12}O_6 + 6 \, O_2$$

7.
$$Zn + Z HCI \rightarrow ZnCl_2 + H_2$$

9.
$$\underline{2}$$
 Na + $\underline{2}$ H₂O \Rightarrow $\underline{2}$ NaOH + $\underline{}$ H₂

12. ____
$$CO_2$$
 + ___ Z $NH_3 \rightarrow ____ $OC(NH_2)_2$ + ____ $H_2O$$

13.
$$_{-}^{4}$$
 Si₂H₃ + $_{-}^{7}$ O₂ \rightarrow $_{-}^{8}$ SiO₂ + $_{-}^{6}$ H₂O₃

14.
$$2 \text{ Al}(OH)_3 + 3 \text{ H}_2SO_4 \rightarrow Al_2(SO_4)_3 + 6 \text{ H}_2O$$

17.
$$2 C_7H_6O_2 + 15 O_2 \rightarrow 14 CO_2 + 5 H_2O$$

19. _____ FeS₂ + _____ O₂
$$\Rightarrow$$
 _____ Fe₂O₃ + _____ SO₂

21. ____ Fe₂O₃ + __3_ H₂
$$\rightarrow$$
 __2_ Fe + __3_ H₂O

22. ____ Na₂CO₃ + __2_ HCl
$$\rightarrow$$
 2 NaCl + ____ H₂O + ___ CO₂

24. ____C₇H₁₆ + _//_O₂
$$\rightarrow$$
 _7 CO₂ + _8 H₂O

25.
$$P_4 + 5 O_2 \rightarrow 2 P_2O_5$$

Writing And Balancing Synthesis Equations Answer Key

The Princeton Review

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Chemistry John McMurry, Robert C. Fay, 2001 A colorful pedagogically enhanced standard textbook for the introductory course It begins with atomic structure proceeds next to bonding and molecules then to bulk physical properties of substances and ends with a study of chemical properties Each chapter concludes with a brief description of an interesting application or extension of the chapter subject a summary a list of key words and a large number of problems Many student oriented supplements are available Annotation copyright by Book News Inc Portland OR Synthesis Green Metrics John Andraos, 2018-12-07 Green chemistry promotes improved syntheses as an intellectual endeavour that can have a great impact both on preserving and utilizing our planet's finite resources and the quality of human life This masterful accomplishment provides an evaluation of environmental impact metrics according to life cycle assessment analysis based on the Mackay compartment environmental model and Guin e environmental impact potentials formalism Assumptions limitations and dealing with missing data are addressed Best literature resources for finding key toxicological parameters are provided and applied to individual reactions as well as entire synthesis plans in order to target molecules of interest Key Features Provides an evaluation of environmental impact metrics according to life cycle assessment analysis Summarises safety hazard metrics according to the same model as life cycle assessment including occupational exposure limits risk phrases flammability and other physical parameters The book will be useful in a range of chemistry courses from undergraduate to advanced graduate courses whether based in lectures tutorials or laboratory experiments Reaction Green Metrics John Andraos, 2018-12-07 This book contains a series of exercises and problems posed in the subject of green metrics Essentially it is a how to book on evaluating the material efficiency environmental impact safety hazard impact and energy efficiency of any kind of chemical reaction or synthesis plan Only the essential green metrics in each of these categories are used The introduction highlights

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