

| Cations \ Anions | +                | - | chloride        | oxide                   | iodide         | hydride        |
|------------------|------------------|---|-----------------|-------------------------|----------------|----------------|
|                  |                  |   | $\text{Cl}^-$   | $\text{O}^{2-}$         | $\text{I}^-$   | $\text{H}^-$   |
| Sodium           | $\text{Na}^+$    |   | $\text{NaCl}$   | $\text{Na}_2\text{O}$   | $\text{NaI}$   | $\text{NaH}$   |
| Potassium        | $\text{K}^+$     |   | $\text{KCl}$    | $\text{K}_2\text{O}$    | $\text{KI}$    | $\text{KH}$    |
| Magnesium        | $\text{Mg}^{2+}$ |   | $\text{MgCl}_2$ | $\text{MgO}$            | $\text{MgI}_2$ | $\text{MgH}_2$ |
| Calcium          | $\text{Ca}^{2+}$ |   | $\text{CaCl}_2$ | $\text{CaO}$            | $\text{CaI}_2$ | $\text{CaH}_2$ |
| Copper(II)       | $\text{Cu}^{2+}$ |   | $\text{CuCl}_2$ | $\text{CuO}$            | $\text{CuI}_2$ | $\text{CuH}_2$ |
| Iron(II)         | $\text{Fe}^{2+}$ |   | $\text{FeCl}_2$ | $\text{FeO}$            | $\text{FeI}_2$ | $\text{FeH}_2$ |
| Iron(III)        | $\text{Fe}^{3+}$ |   | $\text{FeCl}_3$ | $\text{Fe}_2\text{O}_3$ | $\text{FeI}_3$ | $\text{FeH}_3$ |
| Silver           | $\text{Ag}^+$    |   | $\text{AgCl}$   | $\text{Ag}_2\text{O}$   | $\text{AgI}$   | $\text{AgH}$   |
| Zinc             | $\text{Zn}^{2+}$ |   | $\text{ZnCl}_2$ | $\text{ZnO}$            | $\text{ZnI}_2$ | $\text{ZnH}_2$ |
| Aluminum         | $\text{Al}^{3+}$ |   | $\text{AlCl}_3$ | $\text{Al}_2\text{O}_3$ | $\text{AlI}_3$ | $\text{AlH}_3$ |

## Set 2

| Cations \ Anions | +                | - | bromide         | oxide                   | fluoride       | astatide        |
|------------------|------------------|---|-----------------|-------------------------|----------------|-----------------|
|                  |                  |   | $\text{Br}^-$   | $\text{O}^{2-}$         | $\text{F}^-$   | $\text{At}^-$   |
| Lithium          | $\text{Li}^+$    |   | $\text{LiBr}$   | $\text{Li}_2\text{O}$   | $\text{LiF}$   | $\text{LiAt}$   |
| Barium           | $\text{Ba}^{2+}$ |   | $\text{BaBr}_2$ | $\text{BaO}$            | $\text{BaF}_2$ | $\text{BaAt}_2$ |
| Cesium           | $\text{Cs}^+$    |   | $\text{CsBr}$   | $\text{Cs}_2\text{O}$   | $\text{CsF}$   | $\text{CsAt}$   |
| Strontium        | $\text{Sr}^{2+}$ |   | $\text{SrBr}_2$ | $\text{SrO}$            | $\text{SrF}_2$ | $\text{SrAt}_2$ |
| Copper(I)        | $\text{Cu}^+$    |   | $\text{CuBr}$   | $\text{Cu}_2\text{O}$   | $\text{CuF}$   | $\text{CuAt}$   |
| Copper(II)       | $\text{Cu}^{2+}$ |   | $\text{CuBr}_2$ | $\text{CuO}$            | $\text{CuF}_2$ | $\text{CuAt}_2$ |
| Lead(II)         | $\text{Pb}^{2+}$ |   | $\text{PbBr}_2$ | $\text{PbO}$            | $\text{PbF}_2$ | $\text{PbAt}_2$ |
| Lead(IV)         | $\text{Pb}^{4+}$ |   | $\text{PbBr}_4$ | $\text{PbO}_2$          | $\text{PbF}_4$ | $\text{PbAt}_4$ |
| Gallium          | $\text{Ga}^{3+}$ |   | $\text{GaBr}_3$ | $\text{Ga}_2\text{O}_3$ | $\text{GaF}_3$ | $\text{GaAt}_3$ |
| Nickel(II)       | $\text{Ni}^{2+}$ |   | $\text{NiBr}_2$ | $\text{NiO}$            | $\text{NiF}_2$ | $\text{NiAt}_2$ |

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