#### CHAPTER 7 REVIEW

# Chemical Formulas and Chemical Compounds

#### SECTION 7-1

SHORT ANSWER Answer the following questions in the space provided.

- In a Stock name such as iron(III) sulfate, the roman numeral tells us \_\_\_\_\_\_
  - (a) how many atoms of Fe are in one formula unit
  - (b) how many sulfate ions can be attached to the iron atom.
  - (c) the charge on each Fe ion
  - (d) the total positive charge of the formula unit
- The result of changing a subscript in a correctly written chemical formula is to \_\_\_\_\_\_
  - (a) change the number of moles represented by the formula
  - (b) change the charges on the other ions in the compound
  - (c) change the formula so that it no longer represents the compound it previously represented
  - (d) have no effect on the formula
- The explosive TNT has the molecular formula C<sub>2</sub>H<sub>3</sub>(NO<sub>2</sub>)<sub>3</sub>.
  - 4 elements a, How many elements make up this compound?
  - 6 oxygen atoms
    b. How many oxygen atoms are present in one molecule of C<sub>2</sub>H<sub>3</sub>(NO<sub>2</sub>)<sub>3</sub>?
  - 21 atoms c. How many atoms in total are present in one molecule of C<sub>7</sub>H<sub>5</sub>(NO<sub>2</sub>)<sub>3</sub>?
  - 4.2 × 10<sup>24</sup> atoms d. How many atoms are present in a sample of 2 × 10<sup>23</sup> molecules of C<sub>2</sub>H<sub>3</sub>(NO<sub>2</sub>)<sub>3</sub>?
- 4. How many atoms are present in each of these formula units?
  - 11 atoms a. Ca(HCO<sub>3</sub>)<sub>2</sub>
    - 45 atoms b. C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>
    - 10 atoms c. Fe(ClO<sub>2</sub>)<sub>3</sub>
  - 9 atoms d. Fe(ClO<sub>3</sub>)<sub>2</sub>
- 5. N<sub>2</sub>O<sub>5</sub> a. What is the formula for the compound dinitrogen pentoxide?
  - carbon(IV) sulfide b. What is the Stock name for the covalent compound CS2?
    - H<sub>2</sub>SO<sub>3</sub> e. What is the formula for sulfurous acid?
    - phosphoric acid d. What is the name for the acid H<sub>3</sub>PO<sub>4</sub>?

53

# **Writing Chemical Formulas Lab Chapter 7 Answerkey**

**Harold Clark Metcalfe** 

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- 1. Understanding the eBook Writing Chemical Formulas Lab Chapter 7 Answerkey
  - The Rise of Digital Reading Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Writing Chemical Formulas Lab Chapter 7 Answerkey
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Personalized Recommendations
  - Writing Chemical Formulas Lab Chapter 7 Answerkey User Reviews and Ratings
  - Writing Chemical Formulas Lab Chapter 7 Answerkey and Bestseller Lists

- 5. Accessing Writing Chemical Formulas Lab Chapter 7 Answerkey Free and Paid eBooks
  - Writing Chemical Formulas Lab Chapter 7 Answerkey Public Domain eBooks
  - Writing Chemical Formulas Lab Chapter 7 Answerkey eBook Subscription Services
  - Writing Chemical Formulas Lab Chapter 7 Answerkey Budget-Friendly Options
- 6. Navigating Writing Chemical Formulas Lab Chapter 7 Answerkey eBook Formats
  - o ePub, PDF, MOBI, and More
  - Writing Chemical Formulas Lab Chapter 7 Answerkey Compatibility with Devices
  - Writing Chemical Formulas Lab Chapter 7 Answerkey Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Highlighting and Note-Taking Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Interactive Elements Writing Chemical Formulas Lab Chapter 7 Answerkey
- 8. Staying Engaged with Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Writing Chemical Formulas Lab Chapter 7 Answerkey
- 9. Balancing eBooks and Physical Books Writing Chemical Formulas Lab Chapter 7 Answerkey
  - $\circ$  Benefits of a Digital Library
  - Creating a Diverse Reading Collection Writing Chemical Formulas Lab Chapter 7 Answerkey
- 10. Overcoming Reading Challenges
  - o Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Setting Reading Goals Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Fact-Checking eBook Content of Writing Chemical Formulas Lab Chapter 7 Answerkey
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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