



CHEMISTRY

CHEAT SHEET

WRITING IONIC FORMULAS

IONIC COMPOUNDS CHEAT SHEET

WRITING IONIC FORMULAS

How to Write Ionic Formulas

1. Write the symbols for the cation and anion, including their charges.

2. Use the charges to determine the ratio of cations to anions needed to balance the charges.

3. Write the formula, using subscripts to indicate the number of each ion.

4. Check the formula to make sure the charges are balanced.

CATIONS		ANIONS	
1	H^+	1	F^-
2	Li^+	2	O^{2-}
3	Na^+	3	N^{3-}
4	K^+	4	C^{4-}
5	Ag^+	5	P^{5-}
6	Ba^{2+}	6	S^{6-}
7	Ca^{2+}	7	Cl^-
8	Mg^{2+}	8	Br^-
9	Al^{3+}	9	I^-
10	Fe^{2+}	10	At^-
11	Fe^{3+}	11	H^+
12	Cu^+	12	OH^-
13	Cu^{2+}	13	O^{2-}
14	Zn^{2+}	14	F^-
15	Ni^{2+}	15	Cl^-
16	Co^{2+}	16	Br^-
17	Ni^{3+}	17	I^-
18	Co^{3+}	18	At^-
19	Mn^{2+}	19	H^+
20	Mn^{3+}	20	OH^-
21	Mn^{4+}	21	O^{2-}
22	Cr^{2+}	22	F^-
23	Cr^{3+}	23	Cl^-
24	Cr^{6+}	24	Br^-
25	Fe^{2+}	25	I^-
26	Fe^{3+}	26	At^-
27	Fe^{6+}	27	H^+
28	Fe^{7+}	28	OH^-
29	Fe^{8+}	29	O^{2-}
30	Fe^{9+}	30	F^-
31	Fe^{10+}	31	Cl^-
32	Fe^{11+}	32	Br^-
33	Fe^{12+}	33	I^-
34	Fe^{13+}	34	At^-
35	Fe^{14+}	35	H^+
36	Fe^{15+}	36	OH^-
37	Fe^{16+}	37	O^{2-}
38	Fe^{17+}	38	F^-
39	Fe^{18+}	39	Cl^-
40	Fe^{19+}	40	Br^-
41	Fe^{20+}	41	I^-
42	Fe^{21+}	42	At^-
43	Fe^{22+}	43	H^+
44	Fe^{23+}	44	OH^-
45	Fe^{24+}	45	O^{2-}
46	Fe^{25+}	46	F^-
47	Fe^{26+}	47	Cl^-
48	Fe^{27+}	48	Br^-
49	Fe^{28+}	49	I^-
50	Fe^{29+}	50	At^-
51	Fe^{30+}	51	H^+
52	Fe^{31+}	52	OH^-
53	Fe^{32+}	53	O^{2-}
54	Fe^{33+}	54	F^-
55	Fe^{34+}	55	Cl^-
56	Fe^{35+}	56	Br^-
57	Fe^{36+}	57	I^-
58	Fe^{37+}	58	At^-
59	Fe^{38+}	59	H^+
60	Fe^{39+}	60	OH^-
61	Fe^{40+}	61	O^{2-}
62	Fe^{41+}	62	F^-
63	Fe^{42+}	63	Cl^-
64	Fe^{43+}	64	Br^-
65	Fe^{44+}	65	I^-
66	Fe^{45+}	66	At^-
67	Fe^{46+}	67	H^+
68	Fe^{47+}	68	OH^-
69	Fe^{48+}	69	O^{2-}
70	Fe^{49+}	70	F^-
71	Fe^{50+}	71	Cl^-
72	Fe^{51+}	72	Br^-
73	Fe^{52+}	73	I^-
74	Fe^{53+}	74	At^-
75	Fe^{54+}	75	H^+
76	Fe^{55+}	76	OH^-
77	Fe^{56+}	77	O^{2-}
78	Fe^{57+}	78	F^-
79	Fe^{58+}	79	Cl^-
80	Fe^{59+}	80	Br^-
81	Fe^{60+}	81	I^-
82	Fe^{61+}	82	At^-
83	Fe^{62+}	83	H^+
84	Fe^{63+}	84	OH^-
85	Fe^{64+}	85	O^{2-}
86	Fe^{65+}	86	F^-
87	Fe^{66+}	87	Cl^-
88	Fe^{67+}	88	Br^-
89	Fe^{68+}	89	I^-
90	Fe^{69+}	90	At^-
91	Fe^{70+}	91	H^+
92	Fe^{71+}	92	OH^-
93	Fe^{72+}	93	O^{2-}
94	Fe^{73+}	94	F^-
95	Fe^{74+}	95	Cl^-
96	Fe^{75+}	96	Br^-
97	Fe^{76+}	97	I^-
98	Fe^{77+}	98	At^-
99	Fe^{78+}	99	H^+
100	Fe^{79+}	100	OH^-
101	Fe^{80+}	101	O^{2-}
102	Fe^{81+}	102	F^-
103	Fe^{82+}	103	Cl^-
104	Fe^{83+}	104	Br^-
105	Fe^{84+}	105	I^-
106	Fe^{85+}	106	At^-
107	Fe^{86+}	107	H^+
108	Fe^{87+}	108	OH^-
109	Fe^{88+}	109	O^{2-}
110	Fe^{89+}	110	F^-
111	Fe^{90+}	111	Cl^-
112	Fe^{91+}	112	Br^-
113	Fe^{92+}	113	I^-
114	Fe^{93+}	114	At^-
115	Fe^{94+}	115	H^+
116	Fe^{95+}	116	OH^-
117	Fe^{96+}	117	O^{2-}
118	Fe^{97+}	118	F^-
119	Fe^{98+}	119	Cl^-
120	Fe^{99+}	120	Br^-
121	Fe^{100+}	121	I^-
122	Fe^{101+}	122	At^-
123	Fe^{102+}	123	H^+
124	Fe^{103+}	124	OH^-
125	Fe^{104+}	125	O^{2-}
126	Fe^{105+}	126	F^-
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168	Fe^{147+}	168	Br^-
169	Fe^{148+}	169	I^-
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178	Fe^{157+}	178	At^-
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186	Fe^{165+}	186	At^-
187	Fe^{166+}	187	H^+
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230	Fe^{209+}	230	F^-
231	Fe^{210+}	231	Cl^-
232	Fe^{211+}	232	Br^-
233	Fe^{212+}	233	I^-
234	Fe^{213+}	234	At^-
235	Fe^{214+}	235	H^+
236	Fe^{215+}	236	OH^-

Writing Ionic Formulas Study Guide

Martha Mackin



Writing Ionic Formulas Study Guide:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover

Study Guide to Accompany Calculus for the Management, Life, and Social Sciences Clyde Metz, 1984-01-01 Study Guide to Accompany Calculus for the Management Life and Social Sciences

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Basic Concepts of Chemistry, 9e Study Guide

and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

General Chemistry Study Guide Sixth Edition Darrell D. Ebbing, 1999 *Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry* Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts. After a brief review of each section's critical ideas, students are taken through stepped-out worked examples, try it yourself examples, and chapter quizzes, all structured to reinforce chapter objectives and build problem-solving techniques. The solutions manual includes detailed solutions to all odd-numbered exercises in the text.

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Reproduce 9 Heredity and Evolution 10 Light Reflection and Refraction 11 The Human Eye and the Colourful World 12 Electricity 13 Magnetic Effects of Electric Current 14 Sources of Energy 15 Our Environment 16 Sustainable Management of Natural Resources Practical Project Appendix Answer Sheet Examination Paper *Chemistry Equations And Answers (Speedy Study Guides)* Speedy Publishing, 2015-04-24 Chemistry is a difficult subject to fully comprehend with its equations and scientific laws Trying to digest an entire book in one semester is a tough job but with the help of study guides like these you can absorb information in chemistry much more effectively This guide covers chemical equations including examples potential problems and solutions **General Studies Manual Paper-1 2022** Dr. Priya Goel, Jenmenjay Sahni, Vivek Singh, Deepika Singla, Tushar Shukla, Amibh Ranjan, Rakesh Kumar Roshan, Ekta Soni, Sahana Ansari, Dr Poonam Singh, 2021-12-20 1 General Studies Paper 1 is the best selling book particularly designed for the civil services Preliminary examinations 2 This book is divided into 6 major sections covering the complete syllabus as per UPSC pattern 3 Special Section is provided for Current Affairs covering events Summits and Conferences 4 simple and lucid language used for better understanding of concepts 5 5 Crack Sets are given for practice 6 Practice Questions provides Topicwise Questions and Previous Years Solved Papers With our all time best selling edition of General Studies Manual Paper 1 is a guaranteed success package which has been designed to provide the complete coverage to all subjects as per prescribed pattern along with the updated and authentic content The book provides the conventional Subjects like History Geography Polity and General Science that are thoroughly updated along with Chapterwise and Sectionwise questions Contemporary Topics likes Indian Economy Environment Ecology Science Technology and General Awareness have also been explained with latest facts and figures to ease the understanding about the concepts in this book Current events of national and international interest have been listed in a separate section Practice Sets are given at the end keeping in view the trend of the questions coming in exams Lastly More than 5000 Most Important Points for Revision are provided in the attached booklet of the guide It is a must have tool that proves to be one point solution for the preparf Civil Services Preliminary Examination TOC Solved Paper 2021 2018 Indian History and Indian National Movement India and World Geography Indian Polity and Governance Indian Economy General Science Science and Technology General Knowledge Computer Technology Practice Topicwise Questions Current Affairs Crack Sets 1 5 Study Material Science (2023-24 KVS TGT) YCT Expert Team , 2023 24 KVS TGT Study Material Science Solved Papers Practice Book 2024-25 RRB ALP Stage-I & II Science Study Material and Objective Questions YCT Expert , 2024 25 RRB ALP Stage I II Science Study Material and Objective Questions 288 595 E This book covers Physics Chemistry and Biology **Study Guide** Darrell D. Ebbing, Susan M. Schelble, 1998 **Organic Chemistry, 13e Student Study Guide and Solutions Manual** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-05-03 Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text **Study Guide to Accompany Elements of Chemistry, General, Organic, and Biological -- Robert**

Boikess/Kenneth Breslauer/Edward Edelson Robert A. Klein, 1986

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