Balance the following chemical equation:  $Sb_2S_3(s) + HCl(aq) \longrightarrow SbCl_3(s) + H_2S(g)$ 

# QUESTIONS AND PROBLEMS

# 7.3 Equations for Chemical Reactions

LEARNING GOAL Write a balanced chemical equation from the formulas of the reactants and products for a reaction; determine the number of atoms in the reactants and products.

Determine whether each of the following chemical equations is balanced or not balanced:

a. 
$$S(s) + O_2(g) \longrightarrow SO_3(g)$$

b. 
$$2Al(s) + 3Cl_2(g) \longrightarrow 2AlCl_3(s)$$

c. 
$$H_2(g) + O_2(g) \longrightarrow H_2O(g)$$

d. 
$$C_3H_8(g) + 5O_2(g) \xrightarrow{\Delta} 3CO_2(g) + 4H_2O(g)$$

7.28 Determine whether each of the following chemical equations is balanced or not balanced:

a. 
$$PCl_3(s) + Cl_2(g) \longrightarrow PCl_5(s)$$

**b.** 
$$CO(g) + 2H_2(g) \longrightarrow CH_3OH(g)$$

c. 
$$2KClO_3(s) \longrightarrow 2KCl(s) + O_2(g)$$

d. 
$$Mg(s) + N_2(g) \longrightarrow Mg_3N_2(s)$$

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Jiyuan Zhang

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