

Periodic Trends Worksheet

Directions: Use your notes to answer the following questions.

1. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Oxygen < Carbon < Aluminum < Potassium

2. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Neon < Aluminum < Sulfur < Oxygen

3. Why does fluorine have a higher ionization energy than iodine?

Fluorine has nine protons attracting 9 electrons which are much closer to the nucleus compared to iodine. This results in smaller size of Fluorine than iodine making it difficult to remove an electron from the atom.

4. Why do elements in the same family generally have similar properties?

Because they have same number of electrons in the outer shell (valence electrons) which take part in chemical reaction.

5. Indicate whether the following properties increase or decrease from left to right across the periodic table.

- | | |
|--|------------------|
| a. atomic radius (excluding noble gases) | Decreases |
| b. first ionization energy | Increases |
| c. electronegativity | Increases |

6. What trend in atomic radius occurs down a group on the periodic table? What causes this trend?

Atomic radius **decreases** down the group on the periodic table. As we compare the elements down the group, the effective nuclear charge increases, but at the same time the outermost electrons are found in the shell that is farther away from the nucleus. Also the number of screening electrons increases. This results in reduced attraction between the nucleus and the outermost electrons.

7. What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ionization energy **increases** from left to right across the period. The atomic size becomes smaller from left to right. So it becomes harder to remove electron from the atom. Hence the energy required to do so (ionization Energy) increases.

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Tools for High-quality Differentiated Instruction Cindy Strickland, 2007 60 tools that can be used in every grade and subject designed to help teachers reach higher levels of expertise with differentiation instruction for student learning

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life *Handbook of Formative Assessment* Heidi Andrade, Gregory J. Cizek, 2010-04-02 Formative assessment has recently become a focus of renewed research as state and federal policy makers realize that summative assessments have reached a point of diminishing returns as a tool for increasing student achievement Consequently supporters of large scale testing programs are now beginning to consider the potential of formative assessments to improve student achievement The mission of this handbook is to comprehensively profile this burgeoning field of study Written by leading international scholars and practitioners each chapter includes a discussion of key issues that dominate formative assessment policy and practice today as well as those that are likely to affect research and practice in the coming years Key features include Comprehensive nineteen chapters cover all aspects of formative assessment including classroom assessment large scale applications technological applications applications for special needs students K 12 and post secondary applications psychometric considerations case studies and discussion of alternative assessment formats such as portfolios and performance assessments Integrative thoughtful attention is given to the integration of large scale and classroom assessments Practical provides practical guidance on how to conduct formative assessments that generate credible information to guide instruction Global provides perspectives from leading international scholars and practitioners whose expertise spans diverse settings student populations and educational systems Accessible Style although grounded in the latest research the book s style and tone has been carefully crafted to make it accessible to both the textbook and professional markets It will also be a critical reference book for researchers in teacher preparation educational administration and educational policy studies Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

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scale Many of the original ICCAD volunteers were also members of CANDE Computer Aided Network Design a workshop of the IEEE Circuits and System Society In fact it was at a CANDE workshop that Bill McCalla suggested the creation of a conference for the EDA professional Bill later developed the name The World of Chemistry Jones, Jones Staff, 1996

Clay's Handbook of Environmental Health Stephen Battersby, 2013-03 Clay's Handbook of Environmental Health since its first publication in 1933 has provided a definitive guide for the environmental health practitioner or reference for the consultant or student This twentieth edition continues as a first point of reference reviewing the core principles techniques and competencies and then outlining the specialist subjects It has been refocused on the current curriculum of the UK's Chartered Institute of Environmental Health but should also readily suit the generalist or specialist working outside the UK

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