

Matching

Match each item with the correct statement below.

- | | |
|---------------------|----------------|
| A. solvation | E. electrolyte |
| B. weak electrolyte | F. colloid |
| C. aqueous solution | G. surfactant |
| D. solvent | |

- _____ 1. interferes with hydrogen bonding between water molecules
- _____ 2. dissolving medium
- _____ 3. homogeneous mixture of water and dissolved substances
- _____ 4. Solute ions or molecules are surrounded by solvent molecules.
- _____ 5. compound that will conduct current in the liquid state or in aqueous solution
- _____ 6. compound that ionizes incompletely in aqueous solution
- _____ 7. mixture in which particle size averages between 1 nm and 1000 nm

Match each item with the correct statement below.

- | | |
|----------------------|-------------------|
| A. dispersed phase | E. Tyndall effect |
| B. surface tension | F. suspension |
| C. Brownian motion | G. solute |
| D. dispersion medium | H. emulsion |

- _____ 8. inward force tending to minimize surface area of a liquid
- _____ 9. dissolved particle
- _____ 10. mixture in which particle size averages greater than 1000 nm in diameter
- _____ 11. Colloidal particles spread throughout a suspension.
- _____ 12. phenomenon observed when beam of light passes through a colloid
- _____ 13. chaotic movement of colloidal particles
- _____ 14. colloid of a liquid in a liquid

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 15. How does the surface tension of water compare with the surface tensions of most other liquids?
 - A. It is lower.
 - B. It is about the same.
 - C. It is higher.
 - D. It is higher only when a surfactant is added.
- _____ 16. What causes water's low vapor pressure?

A. dispersion forces	C. hydrogen bonding
B. covalent bonding	D. ionic attractions
- _____ 17. What is the shape of the water molecule?

A. linear	B. tetrahedral	C. trigonal planar	D. bent
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- _____ 18. Which of the following is primarily responsible for holding water molecules together in the liquid state?

Water And Aqueous Systems Answers Guide

DJ Losen



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Table of Contents Water And Aqueous Systems Answers Guide

1. Understanding the eBook Water And Aqueous Systems Answers Guide
 - The Rise of Digital Reading Water And Aqueous Systems Answers Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Water And Aqueous Systems Answers Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Water And Aqueous Systems Answers Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Water And Aqueous Systems Answers Guide
 - Personalized Recommendations

- Water And Aqueous Systems Answers Guide User Reviews and Ratings
- Water And Aqueous Systems Answers Guide and Bestseller Lists
- 5. Accessing Water And Aqueous Systems Answers Guide Free and Paid eBooks
 - Water And Aqueous Systems Answers Guide Public Domain eBooks
 - Water And Aqueous Systems Answers Guide eBook Subscription Services
 - Water And Aqueous Systems Answers Guide Budget-Friendly Options
- 6. Navigating Water And Aqueous Systems Answers Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Water And Aqueous Systems Answers Guide Compatibility with Devices
 - Water And Aqueous Systems Answers Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Water And Aqueous Systems Answers Guide
 - Highlighting and Note-Taking Water And Aqueous Systems Answers Guide
 - Interactive Elements Water And Aqueous Systems Answers Guide
- 8. Staying Engaged with Water And Aqueous Systems Answers Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Water And Aqueous Systems Answers Guide
- 9. Balancing eBooks and Physical Books Water And Aqueous Systems Answers Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Water And Aqueous Systems Answers Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Water And Aqueous Systems Answers Guide
 - Setting Reading Goals Water And Aqueous Systems Answers Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Water And Aqueous Systems Answers Guide
 - Fact-Checking eBook Content of Water And Aqueous Systems Answers Guide

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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