

Chapter 9

Aqueous Solutions

Solutions for Practice Problems

Section 9.1 Making Predictions About Solubility

Student Textbook page 335

1. Problem

Decide whether each of the following salts is soluble or insoluble in distilled water. Give reasons for your answer.

- (a) lead(II) chloride, PbCl_2 (a white crystalline powder used in paints)
 (b) zinc oxide, ZnO (a white pigment used in paints, cosmetics, and calamine lotion)
 (c) silver acetate, AgCH_3COO (a whitish powder that is used to help people quit smoking because of the bitter taste it produces)

What Is Required?

Determine if the salts listed will dissolve in water.

What Is Given?

You have the names of the three salts and access to the solubility guidelines.

Plan Your Strategy

Identify the cations and anions in each salt. Refer to the solubility chart on page 334 of the text and determine the guideline solubility of these ions. The solubility of the salt will correspond with the solubility of the ion having the higher guideline (lower number).

Act on Your Strategy

Name	Formula	Ions	Guideline	Ion Solubility	Salt Solubility
lead(II) chloride	PbCl_2	Pb^{2+}	2	insoluble	INSOLUBLE
		Cl^-	3	soluble	
zinc oxide	ZnO	Zn^{2+}	5	soluble	INSOLUBLE
		O^{2-}	2	insoluble	
silver acetate	AgCH_3COO	Ag^+	2	insoluble	SOLUBLE
		CH_3COO^-	1	soluble	

Check Your Solution

Check your results against another reference. These conclusions are correct.

2. Problem

Which of the following compounds are soluble in water? Explain your reasoning for each compound.

- (a) potassium nitrate, KNO_3 (used to manufacture gunpowder)
 (b) lithium carbonate, Li_2CO_3 (used to treat people who suffer from depression)
 (c) lead(II) oxide, PbO (used to make crystal glass)

What Is Required?

Determine if the salts listed will dissolve in water.

Water And Aqueous Systems Practice Problems Answers

Frederick J. Passman



Water And Aqueous Systems Practice Problems Answers:

A Problem-Solving Approach to Aquatic Chemistry James N. Jensen, 2023-01-05 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter's main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text A Problem Solving Approach to Aquatic Chemistry includes Fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria Solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques Acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity Complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry Oxidation reduction equilibria including equilibrium calculations graphical approaches and applications Gas liquid and solid liquid equilibrium with expanded coverage of the effects of global climate change Other topics including chemical kinetics of aquatic systems surface chemistry and integrative case studies For advanced senior undergraduates and first year graduate students in environmental engineering courses A Problem Solving Approach to Aquatic Chemistry serves as an invaluable learning resource on the topic with a variety of helpful learning elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings

The Geomagnetic Field and Life A. Dubrov, 2013-11-11 I am very pleased that my book The Geomagnetic Field and Life is being published in English in the United States Thanks to the initiative of Plenum Press a publishing house that is widely known in all countries I have a great new opportunity to make direct contact with friends throughout the world My book on the geomagnetic field can be regarded as an abstraction whose purpose is to provide a better picture and understanding of the world around us its main driving forces and factors to help us to know ourselves and to proceed further The essence of the abstraction is that in treating the problem I have deliberately ignored the diverse effects of various external factors on living organisms and have confined myself to an analysis of the effect of the GMF This approach allows me to go one step further to draw various conclusions and propose theories that

might bring us closer to a proper understanding of the true nature of the phenomena Philosophers have long been aware that by such abstract thinking we can determine the nature of phenomena more reliably completely and comprehensively penetrate to the very core of the observed effects and perceive the depth of their interrelations Selected Water Resources Abstracts ,1989 **Extrapolation Practice for Ecotoxicological Effect Characterization of Chemicals** Keith R. Solomon,Theo C.M. Brock,Dick De Zwart,Scott D. Dyer,Leo Posthuma,Sean Richards,Hans Sanderson,Paul Sibley,Paul J. van den Brink,2008-05-23 A wide ranging compilation of techniques Extrapolation Practice for Ecotoxicological Effect Characterization of Chemicals describes methods of extrapolation in the framework of ecological risk assessment The book informally known as EXPECT identifies data needs and situations where these extrapolations can be most usefully applied

Nielsen's Food Analysis Laboratory Manual B. Pam Ismail,S. Suzanne Nielsen,2024-06-20 This fourth edition laboratory manual was written to accompany Nielsen's Food Analysis Sixth Edition by the same authors New to this fourth edition of the laboratory manual are three new chapters that complement both the textbook chapters and the laboratory exercises The book again contains four introductory chapters that help prepare students for doing food analysis laboratory exercises The 26 laboratory exercises in the manual cover 24 of the 35 chapters in the textbook Many of the laboratory exercises have multiple sections to cover several methods of analysis for a particular food component or characteristic Most of the laboratory exercises include the following background reading assignment objective principle of method chemicals reagents precautions and waste disposal supplies equipment procedure data and calculations questions and references This laboratory manual is ideal for the laboratory portion of undergraduate courses in food analysis Chemistry John Olmsted,Gregory M. Williams,1997 Textbook outlining concepts of molecular science Additives in Water-Borne Coatings Gerry Davison,Bruce C Lane,2007-10-31 Additives in Water borne Coatings covers both current technology and the future prognosis for the key additives used in water borne coatings today It brings together international experts to provide a comprehensive practical overview of the field its direction and selection of key additives currently employed for in depth treatment of their use behaviour and scope by expert practitioners in those additives *Lake Michigan Mass Balance Study (LMMB) Methods Compendium: Organic and mercury sample analysis techniques* ,1997 *Annual Book of ASTM Standards* ASTM International,American Society for Testing and Materials,2004 Ebook: Chemistry Julia Burdge,2014-10-16 Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more detailed explanation on areas where she knows they have problems With outstanding art a consistent problem solving approach interesting applications woven throughout the chapters and a wide range of end of chapter problems this is a great third edition text **Relaxation Phenomena in Condensed Matter** William T. Coffey,2009-09-09 Russian contributors provide a synthesis of ideas drawn from dielectric magnetic and elastic relaxation Divided into three sections the book commences

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Grinberg, Sonia Rodriguez, 2019-02-21 This handbook is a guide for workers in analytical chemistry who need a starting place for information about a specific instrumental technique It gives a basic introduction to the techniques and provides leading references on the theory and methodology for an instrumental technique This edition thoroughly expands and updates the chapters to include concepts applications and key references from recent literature It also contains a new chapter on process analytical technology Stantec's Water Treatment John C. Crittenden, R. Rhodes Trussell, David W. Hand, Kerry J.

Howe, George Tchobanoglous, 2022-11-08 The updated third edition of the definitive guide to water treatment engineering now with all new online content Stantec's Water Treatment Principles and Design provides comprehensive coverage of the principles theory and practice of water treatment engineering Written by world renowned experts in the field of public water supply this authoritative volume covers all key aspects of water treatment engineering including plant design water chemistry and microbiology water filtration and disinfection residuals management internal corrosion of water conduits regulatory requirements and more The updated third edition of this industry standard reference includes an entirely new chapter on potable reuse the recycling of treated wastewater into the water supply using engineered advanced treatment technologies QR codes embedded throughout the book connect the reader to online resources including case studies and high quality photographs and videos of real world water treatment facilities This edition provides instructors with access to additional resources via a companion website Contains in depth chapters on processes such as coagulation and flocculation sedimentation ion exchange adsorption and gas transfer Details membrane filtration technologies advanced oxidation and potable reuse Addresses ongoing environmental concerns pharmacological agents in the water supply and treatment strategies Describes reverse osmosis applications for brackish groundwater wastewater and other water sources Includes high quality images and illustrations useful appendices tables of chemical properties and design data and more than 450 exercises with worked solutions Stantec's Water Treatment Principles and Design Updated Third Edition remains an indispensable resource for engineers designing or operating water treatment plants and is an essential textbook for students of civil environmental and water resources engineering **Water Pollution** United States. Congress. Senate. Committee on Public Works. Subcommittee on Air and Water Pollution, 1965 Food Properties Handbook M. Shafiur Rahman, 1995-08-31 133 Illustrations and 252 tables make it fast and easy for you to find the information you need This is the first definitive source of data on physical thermal and thermodynamic properties of foods You can solve your problems in food processing preservation process design and control product development stability determination and sensory analysis With this important new book you can access both theoretical and practical data on properties measurement discover how to apply the

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