Worksheet- Reaction Rates

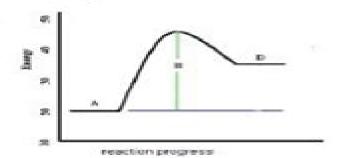
Use this reaction for the questions below: $C_0H_{22}O_0$ (s) + 6 $O_2(g) \rightarrow$ 6 H_2O (g) + 6 CO_2 (g)

- 1. What happens to the concentrations of:
 - a. C₆H₁₂O₆ & O₂ as the reaction proceeds →?
- b. H₂O + CO₂ as the reaction proceeds →?
- 2. According to the collision theory, what 3 circumstances are needed for CoHizOo & Oz to react?
- 3. What is the activation energy for a chemical reaction?
- 4. Use the equation & the collision theory to explain:

$$C_6H_{12}O_6(s) + 6 O_2(g) \rightarrow 6 H_2O(g) + 6 CO_2(g)$$

	Change in condition:	Does this <u>increase</u> or <u>decrease</u> the rate of reaction?	Explain why
a.	Increasing the temperature	Exi (normanca (apenda up)	Ext Molecules more faster to collide more - Increased on rate
ь.	Increasing the concentration of CoHorOs		
C.	Decreasing the concentration of O ₂		
d.	Increase the surface area by chewing up food in your mouth		
6.	Decreasing the temperature		
f.	Increasing the pressure in the container		
9	Decreasing the concentration of H ₂ O		
h.	Increasing the volume of the container the reaction occurs in		
	Increasing the concentration of CO ₂		
1	Using a catalyst (like solivary amylase)		

- On the accompanying energy diagram, <u>label</u> the following terms:
 - a. reactants b. products c. activation energy
- On the graph to the right, <u>draw</u> and <u>label</u> what this diagram would look like if a <u>catalyst</u> was added to the reaction.
- 7. Graph reading
 - a. How much energy (#) do the reactants have?
 - b. How much energy (#) do the products have?
 - c. How much energy (#) is required to activate this un-catalyzed reaction?
 - d. Is this reaction endothermic or exothermic? How do you know?



Reaction Rates And Equilibrium Worksheet

Ralph Thomas Myers

Reaction Rates And Equilibrium Worksheet:

This book delves into Reaction Rates And Equilibrium Worksheet. Reaction Rates And Equilibrium Worksheet is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Reaction Rates And Equilibrium Worksheet, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Reaction Rates And Equilibrium Worksheet
 - Chapter 2: Essential Elements of Reaction Rates And Equilibrium Worksheet
 - Chapter 3: Reaction Rates And Equilibrium Worksheet in Everyday Life
 - Chapter 4: Reaction Rates And Equilibrium Worksheet in Specific Contexts
 - ∘ Chapter 5: Conclusion
- 2. In chapter 1, the author will provide an overview of Reaction Rates And Equilibrium Worksheet. This chapter will explore what Reaction Rates And Equilibrium Worksheet is, why Reaction Rates And Equilibrium Worksheet is vital, and how to effectively learn about Reaction Rates And Equilibrium Worksheet.
- 3. In chapter 2, this book will delve into the foundational concepts of Reaction Rates And Equilibrium Worksheet. The second chapter will elucidate the essential principles that must be understood to grasp Reaction Rates And Equilibrium Worksheet in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Reaction Rates And Equilibrium Worksheet in daily life. The third chapter will showcase real-world examples of how Reaction Rates And Equilibrium Worksheet can be effectively utilized in everyday scenarios.
- 5. In chapter 4, the author will scrutinize the relevance of Reaction Rates And Equilibrium Worksheet in specific contexts. The fourth chapter will explore how Reaction Rates And Equilibrium Worksheet is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, this book will draw a conclusion about Reaction Rates And Equilibrium Worksheet. The final chapter will summarize the key points that have been discussed throughout the book.

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