# TYPES OF CHEMICAL REACTIONS

Identify each type of reaction:

- 1. Na<sub>3</sub> (PO<sub>4</sub>) + 3 KOH --> 3 NaOH + K<sub>3</sub> (PO<sub>4</sub>) double displacement
- MgCl<sub>2</sub> + Li<sub>2</sub>(CO<sub>3</sub>) --> Mg(CO<sub>3</sub>) + 2 LiCl double displacement
- 3. C<sub>6</sub>H<sub>12</sub> + 9 O<sub>2</sub> --> 6 CO<sub>2</sub> + 6 H<sub>2</sub>O combustion
- 4. Pb + Fe(SO<sub>4</sub>) --> Pb(SO<sub>4</sub>) + Fe single displacement
- 5. 2 AgNO<sub>3</sub> + Cu --> Cu(NO<sub>3</sub>)<sub>2</sub> + 2 Ag single displacement
- 6. C<sub>3</sub>H<sub>6</sub>O + 4 O<sub>2</sub> --> 3 CO<sub>2</sub> + 3 H<sub>2</sub>O combustion
- 2 (C<sub>5</sub>H<sub>5</sub>) + Fe --> Fe(C<sub>5</sub>H<sub>5</sub>)<sub>2</sub>
   synthesis
- 8. CaCO<sub>3</sub> --> CaO + CO<sub>2</sub> decomposition
- 9. P<sub>4</sub> + 3 O<sub>2</sub> --> 2 P<sub>2</sub>O<sub>3</sub> synthesis
- 10. 2 RbNO<sub>3</sub> + BeF<sub>2</sub> --> Be(NO<sub>3</sub>)<sub>2</sub> + 2 RbF double displacement

# **Reaction Types Answers Chemistry**

**The Princeton Review** 

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