

CHAPTER 10: Reactions

Reaction Energy**SECTION 1**

Write a balanced chemical equation for the following reactions in the space provided.

- For elements in their standard state, the value of ΔH_f° is _____.
- The formation and decomposition of a substance is represented by the following thermodynamic equations:



- Is the energy being added to or is it being subtracted from the decomposition?
- What is the correct sign for the enthalpy change in this decomposition reaction?

Write the correct answer in the space to the left. Show all your work in the space provided.

- _____ $\Delta H_{\text{f}}^\circ$ of a certain gas is 20 kJ/mol . Absorption of 4 moles of energy would result in the decomposition of _____.

- _____ A compound has a specific heat of $0.001 \text{ J/g}^\circ\text{C}$. If its specific enthalpy is 1 is measured as 100 kJ/mol at 100°C , what is its $\Delta H_{\text{f}}^\circ$ at 100°C ?

- The product is a chemical bond and enthalpy of 100 kJ/mol , and the reactant has an enthalpy of 100 kJ/mol .

- What is the value of ΔH for this reaction?

Reaction Energy Section 1 Answer Key

YCT Expert Team



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Complete Chemistry Rosemarie Gallagher, Rose Marie Gallagher, Paul Ingram, 2000 Complete Chemistry is a revised and enlarged edition of the popular GCSE Chemistry improved to bring it totally up to date This book covers all syllabuses with core material for Double Award and extension material for Science Chemistry The breadth and depth is sufficient to stretch your students aiming for the top grades and makes it an excellent foundation for those intending to progress to advanced level chemistry Key Points Now includes all the necessary topics for IGCSE Concepts and principles of chemistry presented in a clear straightforward style Lively and colourful coverage of the relevance of chemistry in the real world End of chapter testing with more challenging and structured questions Examination style questions Pagination remains the same as GCSE Chemistry so that the two can be used alongside each other

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The Geochemical Origin of Microbes William F. Martin, Karl Kleinermanns, 2024-06-27 This is a textbook covering the transition from energy releasing reactions on the early Earth to energy releasing reactions that fueled growth in the first microbial cells It is for teachers and college students with an interest in microbiology geosciences biochemistry evolution or all of the above The scope of the book is a quantum departure from existing origin of life books in that it starts with basic chemistry and links energy releasing geochemical processes to the reactions of microbial metabolism The text reaches across disciplines providing students of the geosciences an origins biology interface and bringing a geochemistry origins interface to students of microbiology and evolution Beginning with physical chemistry and transitioning across metabolic networks into microbiology the timeline documents chemical events and organizational states in hydrothermal vents the only environments known that bridge the gap between spontaneous chemical reactions that we can still observe in nature today and the physiology of microbes that live from H₂ CO₂ ammonia phosphorus inorganic salts and water Life is a chemical reaction What it is and how it arose are two sides of the same coin Key Features Provides clear connections between geochemical reactions and microbial metabolism Focuses on chemical mechanisms and transition metals Richly illustrated with color figures explaining reactions and processes Covers the origin of the Earth the origin of metabolism the origin of protein synthesis and genetic information as well as the escape into the wild of the first free living cells Bacteria and Archaea

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Chemistry Eileen Ramsden, 2001 This third edition of Key Science Chemistry has been fully revised to meet the requirements of all 2001 GCSE specifications. It is aimed at middle ability students but contains enough material for high achievers. Topics are clearly differentiated between core material for GCSE science Double Award Single Award and extension material for GCSE science chemistry *Fusion Energy Update*, 1980

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