

8.3. Quantitative Phase-Transfer Problems

1. Consider the reaction of the perchlorate of compound (A) in toluene:



It is assumed to proceed via a concerted process in which a molecule of the DMSO-d₆ acts as a catalyst for transfer of the perchlorate.

2. The unimolecular reaction for the rearrangement of compound (B) is shown below:



a. Write the rate law for this reaction.

- b. The initial concentration of the reactants is 0.1 M. The initial concentration of the catalyst is 0.01 M. The initial concentration of the product is 0.01 M. The initial concentration of the reactants is 0.1 M.

- c. The reaction is first order in the reactants and second order in the catalyst. The rate of reaction is 0.1 M/s. The initial concentration of the reactants is 0.1 M. The initial concentration of the catalyst is 0.01 M.

3. The reaction is first order in the reactants and second order in the catalyst.

Reaction Rates And Equilibrium Practice Problems Answer

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