

6. Listed below are the condensed structural formulas or the names for the eight isomers of  $C_5H_{11}Cl$ . Write either formula and the name for each.

1-chloropentane

a.  $CH_3CH_2CH_2CH_2CH_2Cl$

b.  $CH_3CH(CH_3)CH_2CH_2Cl$  1-chloro-3-methylbutane

c. 2-chloropentane

d. 2-chloro-2-methylbutane

e.  $CH_3CH_2CH(Cl)CH_2CH_3$  3-chloropentane

f.  $CH_3CH(CH_3)CH(Cl)CH_3$  2-chloro-3-methylbutane

g. 1-chloro-2-methylbutane

h. 1-chloro-2,2-dimethylpropane

7. Name the following compounds.

a.  $CH_3CH_2CH(CH_3)CH_2CH_2Br$  1-bromo-3-methyl-2-pentylbromide

b.  $CH_2=CHCH(CH_3)CH=CH_2$  3-methyl-1,4-pentadiene

c.  $CH_2=CHCH(CH_3)CH_2CH_3$  2,3-dimethyl-1-pentene

d.  $C_6H_5Cl$  chlorobenzene

e.  $CH_3CH=CHCH_2CH_3$  2-pentene

f.  $CH_3C(CH_3)=CHCH_3$  2,3-dimethyl-2-butene

g.  $CH_3CH_2CH=CH_2$  1-butene

h.  $CH_3CH=CHCH_2CH_3$  2-pentene

i.  $CH_3C(CH_3)=CHCH_2CH_3$  2,3-dimethyl-2-pentene

j.  $CH_2=CHCH(CH_3)CH=CH_2$  2,4-dimethyl-1,4-pentadiene

k.  $CH_2=C(CH_3)C(CH_3)_2CH=CHCH_3$  2,3,3-trimethyl-1,4-hexadiene



8. Draw structural formulas for the following.

a. 3-heptene



b. 2-methylnaphthalene

c. trichloromethane



d. 2-chloro-3-phenylhexane



e. 1,3-cyclopentadiene



f. toluene (methylbenzene)



g. 1,4-dibromobenzene



h. 2-bromo-3-methyl-2-butene



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