

Answer ALL questions. Write your answers in the spaces provided.

Some questions must be answered with a cross in a box ☒. If you change your mind about an answer, put a line through the box ☒ and then mark your new answer with a cross ☒.

- 1 Alumina is the common name for aluminium oxide (Al_2O_3).

Aluminium oxide is the raw material used for the manufacture of aluminium.

- (a) Identify the process used to extract aluminium from aluminium oxide.

(1)

- ☐ A catalysis
☐ B combustion
☒ C electrolysis
☐ D hydration

- (b) Most metal oxides are bases but aluminium oxide is amphoteric.

- (i) Complete the ionic equation for the reaction of aluminium oxide with hydrochloric acid.

(2)



- (ii) Give the missing words A and B in Paragraph 1.

(2)

When aluminium oxide is reacted with hydrogen ions, the aluminium oxide dissolves because it acts as a base.

However, when aluminium oxide reacts with _____ A _____ ions, the aluminium oxide dissolves because it acts as an _____ B _____.

This shows that aluminium oxide is amphoteric.

Paragraph 1

A hydroxide (OH^-)

B acid



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