

PACING GUIDE: CHEMISTRY ESSENTIAL STANDARDS

OBJECTIVES	DAYS	MAJOR TOPICS	LABS/ACTIVITIES	DIGITAL RESOURCES
UNIT 1 – BASICS AND NUMBERS Chem.1.1.1 Analyze the structure of atoms, isotopes, and ions. Chem.2.2.2 Analyze the evidence of chemical change	8	Basics (mixture/substance, physical vs. chemical, metric prefixes, etc.), Sig figs, phases of matter (Note: You can teach the mole here or in two other places noted in this pacing guide.)	*M&M average mass activity *Measurement lab *Demo: Sig figs/precision *Demo: Rxns	*www.teachersdomain.org *Youtube videos
Chem.1.2.3 Compare inter- and intra-particle forces. Chem.1.2.5 Compare the properties of ionic, covalent, metallic, and network compounds. Chem.2.1.1 Explain the energetic nature of phase changes Chem.2.1.2 Explain heating and cooling curves (heat of fusion, heat of vaporization, heat, melting point, and boiling point). Chem.2.1.3 Interpret the data presented in phase diagrams.	3	(Note: Solids and liquids can be included here or in the next unit.)	*"If You Were" phase change game *Phase diagrams *Cooling/heating curves *Surface tension *Demo: water/mason jar	*Animation of phase changes *Actual substance diagrams *Water bug youtube video
UNIT 2 – ATOMS AND THE PERIODIC TABLE Chem.1.1.1 Analyze the structure of atoms, isotopes, and ions. Chem.1.1.2 Analyze an atom in terms of the location of electrons. Chem.1.1.4 Explain the process of radioactive decay by the use of nuclear equations and half-life.	5	Atomic structure, atomic number, mass number isotopes and ions radioactivity including fusion and fission History of the atom can be used for honors enrichment.	*M&M average mass activity	*Youtube – radioactive decay, Gold Foil Exp. *PHET Colorado.edu
See Unit 1	(2)	The Mole		
Chem.1.1.2 Analyze an atom in terms of the location of electrons. Chem.1.1.3 Explain the emission of electromagnetic radiation in spectral form in terms of the Bohr model.	5	Electron configuration, Lewis dot diagrams, nuclear/hyphen form, wave model, quantum numbers	*Write full electron configurations *Hydrogen emission *Flame test *Spectroscopy	*SAS Electron Configuration Activity

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