

Answer ALL questions. Write your answers in the spaces provided.

Some questions must be answered with a cross in a box ☒. If you change your mind about an answer, put a line through the box ☒ and then mark your new answer with a cross ☒.

- 1 Alumina is the common name for aluminium oxide (Al_2O_3).

Aluminium oxide is the raw material used for the manufacture of aluminium.

- (a) Identify the process used to extract aluminium from aluminium oxide.

(1)

- ☐ A catalysis
☐ B combustion
☒ C electrolysis
☐ D hydration

- (b) Most metal oxides are bases but aluminium oxide is amphoteric.

- (i) Complete the ionic equation for the reaction of aluminium oxide with hydrochloric acid.

(2)



- (ii) Give the missing words A and B in Paragraph 1.

(2)

When aluminium oxide is reacted with hydrogen ions, the aluminium oxide dissolves because it acts as a base.

However, when aluminium oxide reacts with _____ A _____ ions, the aluminium oxide dissolves because it acts as an _____ B _____.

This shows that aluminium oxide is amphoteric.

Paragraph 1

A hydroxide (OH^-)

B acid



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Table of Contents Paljas Past Paper Questions And Answers

1. Understanding the eBook Paljas Past Paper Questions And Answers
 - The Rise of Digital Reading Paljas Past Paper Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Paljas Past Paper Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Paljas Past Paper Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Paljas Past Paper Questions And Answers
 - Personalized Recommendations
 - Paljas Past Paper Questions And Answers User Reviews and Ratings
 - Paljas Past Paper Questions And Answers and Bestseller Lists
5. Accessing Paljas Past Paper Questions And Answers Free and Paid eBooks
 - Paljas Past Paper Questions And Answers Public Domain eBooks
 - Paljas Past Paper Questions And Answers eBook Subscription Services
 - Paljas Past Paper Questions And Answers Budget-Friendly Options
6. Navigating Paljas Past Paper Questions And Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Paljas Past Paper Questions And Answers Compatibility with Devices
 - Paljas Past Paper Questions And Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Paljas Past Paper Questions And Answers
 - Highlighting and Note-Taking Paljas Past Paper Questions And Answers
 - Interactive Elements Paljas Past Paper Questions And Answers
8. Staying Engaged with Paljas Past Paper Questions And Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Paljas Past Paper Questions And Answers
- 9. Balancing eBooks and Physical Books Paljas Past Paper Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Paljas Past Paper Questions And Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Paljas Past Paper Questions And Answers
 - Setting Reading Goals Paljas Past Paper Questions And Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Paljas Past Paper Questions And Answers
 - Fact-Checking eBook Content of Paljas Past Paper Questions And Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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