

## Naming Compounds Containing Polyatomic Ions

Use the following list of polyatomic ions to name the compounds given.

OH  
NO<sub>3</sub>  
C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>

hydroxide  
nitrate  
acetate

ClO<sub>3</sub>  
HCO<sub>3</sub>  
SO<sub>4</sub>  
CO<sub>3</sub>

chlorate  
bicarbonate  
sulfate  
carbonate

SO<sub>3</sub>  
PO<sub>4</sub>  
NH<sub>4</sub>

sulfite  
phosphate  
ammonium

- |  |   |
|--|---|
| 1. CuSO <sub>4</sub> _____                             | 26. NaOH _____  |
| 2. KClO <sub>3</sub> _____                             | 27. NaNO <sub>3</sub> _____                               |
| 3. NH <sub>4</sub> OH _____                            | 28. Ba(OH) <sub>2</sub> _____                             |
| 4. K <sub>2</sub> CO <sub>3</sub> _____                | 29. K <sub>2</sub> CO <sub>3</sub> _____                  |
| 5. Na <sub>2</sub> SO <sub>4</sub> _____               | 30. LiOH _____  |
| 6. KC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> _____ | 31. KHCO <sub>3</sub> _____                               |
| 7. NH <sub>4</sub> Cl _____                            | 32. NH <sub>4</sub> F _____                               |
| 8. Mg(OH) <sub>2</sub> _____                           | 33. KClO <sub>3</sub> _____                               |
| 9. Zn(NO <sub>3</sub> ) <sub>2</sub> _____             | 34. NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> _____  |
| 10. MgSO <sub>4</sub> _____                            | 35. CaClO <sub>3</sub> _____                              |
| 11. CaCO <sub>3</sub> _____                            | 36. SrNO <sub>3</sub> _____                               |
| 12. KOH _____  | 37. Be(OH) <sub>2</sub> _____                             |
| 13. Ca(OH) <sub>2</sub> _____                          | 38. Li <sub>2</sub> SO <sub>4</sub> _____                 |
| 14. K <sub>2</sub> CO <sub>3</sub> _____               | 39. Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> _____ |
| 15. AlPO <sub>4</sub> _____                            | 40. BaSO <sub>3</sub> _____                               |
| 16. Na <sub>2</sub> PO <sub>4</sub> _____              | 41. NH <sub>4</sub> I _____                               |
| 17. MgSO <sub>4</sub> _____                            | 42. NH <sub>4</sub> HCO <sub>3</sub> _____                |
| 18. AgNO <sub>3</sub> _____                            | 43. Cu <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub> _____ |
| 19. Na <sub>2</sub> CO <sub>3</sub> _____              | 44. Ni(OH) <sub>2</sub> _____                             |
| 20. NaHCO <sub>3</sub> _____                           | 45. ZnSO <sub>4</sub> _____                               |
| 21. Ca(NO <sub>3</sub> ) <sub>2</sub> _____            | 46. MnCO <sub>3</sub> _____                               |
| 22. BaSO <sub>4</sub> _____                            | 47. H <sub>2</sub> SO <sub>4</sub> _____                  |
| 23. BPO <sub>3</sub> _____                             | 48. Au <sub>2</sub> SO <sub>4</sub> _____                 |
| 24. LiNO <sub>3</sub> _____                            | 49. NH <sub>4</sub> NO <sub>3</sub> _____                 |
| 25. Al(NO <sub>3</sub> ) <sub>3</sub> _____            | 50. LiClO <sub>3</sub> _____                              |

# Polyatomic Ions Naming Answer

**Richard Post, Chad Snyder, Clifford C.  
Houk**



## **Polyatomic Ions Naming Answer:**

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students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book s unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink [http://www.bfwpub.com/pdfs/wink/POCPowerPoint\\_Final.ppt](http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt) 832KB

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can make or break a career      **Foundations of College Chemistry, Alternate** Morris Hein, Susan Arena, 2010-01-26

Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis      **Survival Guide to General Chemistry**

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Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit [www.whfreeman.com/gob](http://www.whfreeman.com/gob)      **Introductory Chemistry** Michael P.

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focused active classroom that range from cooperative learning to active student participation in a more traditional setting

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**Let's Review Regents: Chemistry--Physical Setting Revised Edition** Barron's Educational Series, Albert S. Tarendash, 2021-01-05 Barron's Let's Review Regents Chemistry gives students the step by step review and practice they need to prepare for the Regents Chemistry Physical Setting exam This updated edition is an ideal companion to high school textbooks and covers all Chemistry topics prescribed by the New York State Board of Regents Let's Review Regents Chemistry covers all high school level Chemistry topics and includes Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key **Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced eText with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order **eBook:**

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Mikulecky, 2022-11-23 Everything you need to crush chemistry with confidence Chemistry All in One For Dummies arms you with all the no nonsense how to content you'll need to pass your chemistry class with flying colors You'll find tons of practical examples and practice problems and you'll get access to an online quiz for every chapter Reinforce the concepts you learn in the classroom and beef up your understanding of all the chemistry topics covered in the standard curriculum Prepping for



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