

5.1 Knowledge preview

FOUNDATION

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- 1 In the space below draw a simple labelled diagram to show what you know about the parts of an atom.

There are 3 parts to an atom: protons, neutrons and electrons.

- Protons and neutrons are both located in the nucleus in the centre of the atom and are relatively large.
- Electrons are located in the electron cloud (or in electron shells) and are much smaller.
- Protons have a charge of +1, electrons have a charge of -1 and neutrons have no charge.

- 2 List the names and symbols for 10 common elements in the periodic table.

Individual response

- 3 Define the following terms:

(a) atom the smallest particle of matter

(b) element a substance made of only 1 type of atoms, for example oxygen gas O_2 is made of only oxygen atoms

(c) compound a molecule made of more than 1 type of atom, for example water H_2O is made of two hydrogen atoms and one oxygen atom bonded together

(d) crystal lattice a large regular arrangement of particles. The particles can be ions like sodium and chloride ions in table salt, molecules, or atoms of a single element like diamond

- 4 (a) Compare an atom and an ion, listing their similarities and differences.

An ion is an atom (or group of atoms) that has gained or lost electrons so that it has become charged

(b) What is the difference between a positive ion (cation) and a negative ion (anion)? In your answer, include how each ion forms.

Positive ions have lost electrons to become positively charged whereas negative ions have gained electrons to become negatively charged

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