

Chemical Equations and Reactions

SECTION 2

SHORT ANSWER Answer the following questions in the space provided.

1. Match the equation type on the left to its representation on the right.

- | | |
|----------------------------------|-----------------------------------|
| <u> c </u> synthesis | (a) $AX + BY \rightarrow AY + BX$ |
| <u> d </u> decomposition | (b) $A + BX \rightarrow AX + B$ |
| <u> b </u> single-displacement | (c) $A + B \rightarrow AX$ |
| <u> a </u> double-displacement | (d) $AX \rightarrow A + X$ |

2. c In the reaction described by the equation $2Al(s) + 3Fe(NO_3)_2(aq) \rightarrow 3Fe(s) + 2Al(NO_3)_3(aq)$, iron has been replaced by

- (a) nitrate. (c) aluminum.
(b) water. (d) nitrogen.

3. a Of the following chemical equations, the only reaction that is both synthesis and combustion is

- (a) $C(s) + O_2(g) \rightarrow CO_2(g)$.
(b) $2C_4H_{10}(l) + 13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(l)$.
(c) $6CO_2(g) + 6H_2O(g) \rightarrow C_6H_{12}O_6(aq) + 6O_2(g)$.
(d) $C_6H_{12}O_6(aq) + 6O_2(g) \rightarrow 6CO_2(aq) + 6H_2O(l)$.

4. b Of the following chemical equations, the only reaction that is both combustion and decomposition is

- (a) $S(s) + O_2(g) \rightarrow SO_2(g)$.
(b) $2C_4H_{10}(l) + 13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(l)$.
(c) $2H_2O_2(l) \rightarrow 2H_2O(l) + O_2(g)$.
(d)

5. Identify the products when the following substances decompose:

- a. **Its separate elements** _____ a binary compound
b. **Metal oxide + water** _____ most metal hydroxides
c. **Metal oxide + carbon dioxide** _____ a metal carbonate
d. **Water + sulfur dioxide** _____ the acid H_2SO_3

6. The complete combustion of a hydrocarbon in excess oxygen yields the products

Review Chemical Equations And Reactions Section 3

Thomas Elliott Taylor



Review Chemical Equations And Reactions Section 3:

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An Introduction to Chemistry Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

Combustion Science and Engineering Kalyan Annamalai, Ishwar K. Puri, 2006-12-19 While the basic concepts in any discipline may not vary the problems it faces do change Combustion Science and Engineering is for students and engineers who wish to understand combustion fundamentals and apply them to engineering problems In several instances the authors have included physical explanations along with the mathematical relations and equations so that the principles can be applied to solve real world combustion and pollution problems The book contains an outline of the corpuscular aspects of thermodynamics and introduces the background related to combustion of solid liquid and gaseous fuels Exercise problems formulae and tables appear at the end of text Students embarking on their studies in chemical mechanical aerospace energy and environmental engineering will face continually changing combustion problems such as pollution control and energy efficiency throughout their careers

Sif: Chemistry 5na Tb J. G. R. Briggs, 2009

Basic Chemical Concepts and Tables Steven L. Hoenig, 2024-05-16 Fully revised and expanded the second edition of Basic Chemical Concepts and Tables is written as a quick reference to the many different concepts and ideas encountered in chemistry The volume presents important subjects in a concise format that makes it a practical resource for any reader Subjects include general chemistry inorganic chemistry organic chemistry and spectral analysis The new edition includes updated tables that are useful for the interpretation of ultraviolet visible UV Vis infrared IR nuclear magnetic resonance NMR and mass spectroscopy MS spectra and expanded sections devoted to the concept of isomers and polymer structures and includes a new chapter on nuclear chemistry Separate chapters offer physical constants and unit measurements commonly encountered and mathematical concepts needed when reviewing or working with basic chemistry concepts Key features Provides chemical information in a concise format fully illustrated with many graphs and

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Chemistry: Core Concepts, 3rd Edition Allan Blackman, Daniel Southam, Gwendolyn Lawrie, Natalie Williamson, Christopher Thompson, 2024-01-09 The third edition of Chemistry Core Concepts Blackman et al has been developed by a group of leading chemistry educators for students entering university with little or no background in chemistry Available as a full colour printed textbook with an interactive eBook code this title enables every student to master concepts and succeed in assessment Lecturers are supported with an extensive and easy to use teaching and learning package *Cehmistry Textbook for College and University USA* Ibrahim Sikder, 2023-06-04 Cehmistry Textbook USA

Chemical Reactions and Chemical Reactors George W. Roberts, 2008-03-14 Focused on the undergraduate audience Chemical Reaction Engineering provides students with complete coverage of the fundamentals including in depth coverage of chemical kinetics By introducing heterogeneous catalysis early in the book the text gives students the knowledge they need to solve real chemistry and industrial problems An emphasis on problem solving and numerical techniques ensures students learn and practice the skills they will need later on whether for industry or graduate work **Basic Concepts of**

Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Introduction to Chemical Reactor Analysis R.E. Hayes, 2020-12-17 This

book provides an introduction to the basic concepts of chemical reactor analysis and design It is intended for both the senior level undergraduate student in chemical engineering and the working professional who may require an understanding of the basics of this subject **Biochemistry** Donald Voet, Judith G. Voet, 2010-12-01 The Gold Standard in Biochemistry text books Biochemistry 4e is a modern classic that has been thoroughly revised Don and Judy Voet explain biochemical concepts

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