#### Chapter 10 Nuclear Chemistry

- How is the product isotope different from the reactant isotope in alpha decay? The product isotope has two fewer protons and two fewer neutrons than the reactant isotope.
- Circle the letters that identify each event that takes place during beta decay.
  - a. A proton decomposes into a neutron and an electron.
  - (b) A neutron decomposes into a proton and an electron.
  - c.) An electron is emitted from the nucleus.
  - A neutron is emitted from the nucleus.
- Why are beta particles more penetrating than alpha particles?Beta particles have a smaller mass and a faster speed.
- 10. Is the following sentence true or false? All nuclear radiation consists of charged particles.
- What is a gamma ray? A gamma ray is a penetrating ray of energy emitted by an unstable nucleus.
- 12. How fast do gamma rays travel through space? Gamma rays travel through space at the speed of light.
- 13. Complete the following table about nuclear radiation.

Characteristics of Nuclear Radiation			
Radiation Type	Charge	Mass (amu)	Usually Stopped By
Alpha particle	-1	4	Paper or clothing
Beta particle	1-	1 1836	Aluminum sheet
Garmena raiy	0	0	Several meters of concrete

#### Effects of Nuclear Radiation (pages 296-297)

- 14. How does nuclear radiation affect atoms? Nuclear radiation can ionize atoms.
- Is the following sentence true or false? One potential danger of radon gas is that prolonged exposure to it can lead to lung cancer.

#### Detecting Nuclear Radiation (page 297)

- Name two devices that are used to detect nuclear radiation.
  - а, Geiger counters b, Film badges

# **Review Nuclear Chemistry Section 2 Answer Key**

**D** Siedentop

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to practitioners and policymakers involved in environmental management and public policy Actinides and the **Environment** P.A. Sterne, A. Gonis, A.A. Borovoi, 1998-02-28 The handling of actinides and actinide based materials provides significant technological challenges due to the toxicity and radioactivity associated with these materials These challenges are particularly apparent in the nuclear power industry Under normal operation a reactor can produce a significant amount of spent fuel requiring subsequent containment for geologic times and under accident conditions it can release lethal doses of radioactive material to the environment Inevitably radioactive material will enter the environment necessitating as complete an understanding as possible of its behavior An understanding of the interaction between actinides and the environment must be based on a knowledge of their basic physical and chemical properties To date although there is general agreement on the principles for waste disposal no facility has been built for the long term disposal of high level radioactive waste from either normal reactor operations or from accidental catastrophes This makes it most important for the scientific and technical community to develop the necessary cross disciplinary understanding that will help us implement safe and secure waste management accident remediation and accident prevention systems **Resources in Education**, 1996 Title List of Documents Made Publicly Available U.S. Nuclear Regulatory Commission, 1992 Publications of the National Bureau of Standards ... Catalog United States. National Bureau of Standards, 1969 The Compu-mark Directory of U.S. Publications United States. National Bureau of Standards, 1969 **Cehmistry Textbook for College** Trademarks ,1991 and University USA Ibrahim Sikder, 2023-06-04 Cehmistry Textbook USA **Publications of the National Bureau of Standards** United States. National Bureau of Standards, 1968 Parliamentary Debates (Hansard). Great Britain. Isotopes and Radiation Technology ,1965 Parliament. House of Commons, 1997 Essentials of Physical Chemistry Don Shillady, 2011-07-27 At a time when U S high school students are producing low scores in mathematics and science on international examinations a thorough grounding in physical chemistry should not be considered optional for science undergraduates Based on the author's thirty years of teaching Essentials of Physical Chemistry merges coverage of calculus with chemistry and molecular physics in a friendly yet thorough manner Reflecting the latest ACS guidelines the book can be used as a one or two semester course and includes special topics suitable for senior projects The book begins with a math and physics review to ensure all students start on the same level and then discusses the basics of thermodynamics and kinetics with mathematics tuned to a level that stretches students abilities It then provides material for an optional second semester course that shows students how to apply their enhanced mathematical skills in a brief historical development of the quantum mechanics of molecules Emphasizing spectroscopy the text is built on a foundation of quantum chemistry and more mathematical detail and examples It contains sample classroom tested exams to gauge how well students know how to use relevant formulas and to display successful understanding of key concepts Coupling the development of mathematical skills with chemistry concepts encourages students to learn mathematical derivations Mini biographies of famous scientists make

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