Name	Date	Class
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## 7-3 Review and Reinforcement

#### Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

Number	Prefix	
1	mono	
2	di	
3	tri	
4	tetra	
5	penta	
6	hexa	
7	hepta	1-000
8	octa	
9	nona	
10	deca	

Complete each of the following sentences by filling in the appropriate word or phrase from the list below

cation hydrates electronegative ionic anion anhydrates hydrogen ions molecular

11. Ionic compounds that absorb water into their solid structures form substances called

hydrate —

- 12. When naming an ionic substance, the name of the \_\_cation \_\_\_\_\_ comes first
- To name a binary molecular compound, add the suffix -ide to the more\_electronegative
  of the two elements.
- 14. By one definition, an acid is a molecular compound that dissolves in water to produce hydrogen ions
- 15. The name of an acid is usually derived from the name of its characteristic anion
- 16. Prefixes are often used when naming molecular compounds.

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John S. Phillips, Cheryl Wistrom

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Clinical Chemistry - E-Book Donna Larson, 2015-12-17 Gain a clear understanding of pathophysiology and lab testing Clinical Chemistry Fundamentals and Laboratory Techniques prepares you for success as a medical lab technician by simplifying complex chemistry concepts and lab essentials including immunoassays molecular diagnostics and quality control A pathophysiologic approach covers diseases that are commonly diagnosed through chemical tests broken down by body system and category such as respiratory gastrointestinal and cardiovascular conditions Written by clinical chemistry educator Donna Larson and a team of expert contributors this full color book is ideal for readers who may have minimal knowledge of chemistry and are learning laboratory science for the first time Full color illustrations and design simplify complex concepts and make learning easier by highlighting important material Case studies help you apply information to real life scenarios Pathophysiology and Analytes section includes information related to diseases or conditions such as a biochemistry review disease mechanisms clinical correlation and laboratory analytes and assays Evolve companion website includes case studies and animations that reinforce what you ve learned from the book Laboratory Principles section covers safety quality assurance and other fundamentals of laboratory techniques Review questions at the end of each chapter are tied to the learning objectives helping you review and retain the material Critical thinking questions and discussion questions help you think about and apply key points and concepts Other Aspects of Clinical Chemistry section covers therapeutic drug monitoring toxicology transplantation and emergency preparedness Learning objectives in each chapter help you to remember key points or to analyze and synthesize concepts in clinical chemistry A list of key words Is provided at the beginning of each chapter and these are also bolded in the text Chapter summaries consist of bulleted lists and tables highlighting the most important points of each chapter A glossary at the back of the book provides a quick reference to definitions of all clinical chemistry terms **Environmental Science and Technology** Diana L. Turner, 2003

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