Write increases or decreases to complete each relationship.

- As volume increases, pressure _____decreases______
- 2. As volume increases, temperature increases
- As pressure increases, temperature _____increases______.

True or False.

- 6. ___T___ The Kinetic Theory of Matter deals with moving particles.
- 7. ____T___ Gases can be collected by trapping them over water.
- T Real gases exist in the world around us.
- T_____ Densities of gases are lower than 1 g/mL.
- F_____ If the barometric pressure is falling, that means good weather is coming.
- 11. ____T __ The troposphere contains the most air particles.
- T_____T Temperature is a measure of how fast particles are moving.
- 13. The gas that makes up the highest percentage of the atmosphere is

____nitrogen_____.

- 14. Atmospheric pressure is measured using a barometer .
- 15. Squeezing gas particles together is known as _____compressibility_______.
- 16. When particles collide with the sides of a container, they create pressure .
- 17. The spreading of a gas through a small opening is _____effusion_____
- 18. The number of gas particles present is called the ______concentration______.

Gas Law Problems. Write the answer with the correct unit.

- A gas has a beginning pressure of 2.95 atm and a volume of 5.15 L. If the ending pressure is 4.23 atm, what is the ending volume? P₂V₂ = P₂V₂
 (2.95)(5.15) ÷ 4.23 = 3.59
- 20. A gas has a beginning temperature of 300 K and a volume of 4.06 L. If the ending temperature is 176 K, what is the ending volume? V₂T₂ = V₂T₃
 (4.06)(176) ÷ 300 = 2.38
- 21. A gas has a beginning pressure of 2.09 atm, a temperature of 268 K, and a volume of 0.450 L. If the ending pressure is 1.2 atm and the temperature is 149 K, what is the ending volume? P₃V₃T₂ = P₂V₂T₃

Review Section 1 Gases Answers

JL Elias

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